# On the mechanism of chloroquine resistance in Plasmodium falciparum 

Supporting Information

Mauro Chinappi ${ }^{1^{*}}$, Allegra Via ${ }^{1^{*}}$, Paolo Marcatili ${ }^{1}$ and Anna Tramontano ${ }^{1,2,+}$<br>${ }^{1}$ Department of Biochemical Sciences, Sapienza University, Rome, Italy<br>${ }^{2}$ Istituto Pasteur, Fondazione Cenci Bolognetti, Sapienza University, Rome, Italy

## Table of Contents

S1-PfCRT cannot be a channel for un-protonated chloroquine CQ
S2 - Membrane equilibrium equation
S3 - The expression for the Cellular Accumulation Ratio of chloroquine
S4 - Chloroquine concentration is equal in the external buffer and inside the erythrocyte
S5 - CAR expressions in the case of chloroquine-HM binding in linear regime (hypothesis H1)
S6 - Proof that $[C Q]_{D V, D}=[C Q]_{\mathrm{e}}$ in the hypothesis PfCRT is a channel for the protonated forms of chloroquine
S7 - Value of the channel permeability in the hypothesis PfCRT is a channel and $[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}}=$ $\alpha[C Q]_{D v}$
S8 - Calculation of $\mathrm{CAR}_{\mathrm{D}}$ in the hypothesis [CQ:HM $]_{\mathrm{DV}}=\alpha[\mathrm{CQ}]_{\mathrm{Dv}}(\mathrm{H} 1)$ and PfCRT is a non saturated active carrier (J2a) (main text, Figure 1, cells C3-5,5)
S9 - Detailed numerical calculation showing that chloroquine-HM binding in experiment C is not in the saturation region
S10 - Proof that eq. (18) (main text) is an increasing function of $\left[\mathrm{H}^{+}\right]_{\mathrm{DV}}$ in the interval $0<\mathrm{pH}_{\mathrm{DV}}<9.15$
S 12 - Calculation of $\mathrm{CAR}_{\mathrm{D}}$ in saturated condition for chloroquine-HM binding and $\mathrm{J}_{\mathrm{PfCRT}}=\lambda\left[\mathrm{CQ}^{++}\right]_{\mathrm{DV}}$ (cells C5,1-4, Figure 1 of the main text])
S13-Calculations relating to each single hypothesis for the chloroquine:HM binding in saturated condition and $\mathrm{J}_{\mathrm{PfCRT}}=\lambda\left[\mathrm{CQ}^{+}\right]_{\mathrm{DV}}$ (cells C4,1-4, Figure 1 of the main text $]$ )
S14-Prediction for $\mathrm{CAR}_{\mathrm{D}}$ in saturated condition for chloroquine-HM binding (H2) and $\mathrm{J}_{\mathrm{PfCRT}}=$ const (J2b) (cells C2,1-4,Figure 1, main text)

## S1-PfCRT cannot be a channel for un-protonated chloroquine CQ

At equilibrium, the net flux of chloroquine across the vacuolar membrane is zero. Let $\mathrm{J}_{\text {mem }}$ be the inward membrane flux of unprotonated chloroquine CQ and $\mathrm{J}_{\text {PfCRT }}$ the chloroquine outward flux due to the CQR mutated form of PfCRT. We have:

$$
\begin{equation*}
J_{m e m}=J_{P f C R T} \tag{eq-S1}
\end{equation*}
$$

[^0]In the hypothesis that PfCRT is a passive channel for CQ, both $\mathrm{J}_{\text {mem }}$ and $\mathrm{J}_{\text {PfCRT }}$ are only functions of the difference of CQ concentrations on the two sides of the vacuolar membrane, i.e.

$$
\begin{equation*}
g\left([C Q]_{e}-[C Q]_{D V}\right)=f\left([C Q]_{D V}-[C Q]_{e}\right) \tag{eq-S2}
\end{equation*}
$$

where square brackets indicate concentrations and $f(x)$ and $g(x)$ are increasing functions of their arguments that are equal to zero when the argument is zero. The only possible solution of (eq-S2) is $[C Q]_{\mathrm{DV}}=[\mathrm{CQ}]_{\mathrm{e}}$ (the analytical proof is conceptually identical to the one reported below in section S6). Notice that, in CQS strains, where PfCRT is not involved, the membrane equilibrium (expressed as: $\left.\mathrm{g}\left([\mathrm{CQ}]_{\mathrm{e}}-[\mathrm{CQ}]_{\mathrm{DV}}\right)=0\right)$ implies $[\mathrm{CQ}]_{\mathrm{Dv}}=[C Q]_{\mathrm{e}}$ as well. Therefore, the hypothesis that PfCRT is a channel for CQ entails that unprotonated chloroquine accumulation in the vacuole of $C Q R$ and $C Q S$ strains is the same. Having CQS and CQR strains identical $\mathrm{pH}_{\mathrm{DV}}$ values [1] [2] [3], their vacuole must also have identical concentration of $\mathrm{CQ}^{+}$and $\mathrm{CQ}^{++}$. As a consequence, the two strains also have the same value of $[\mathrm{CQ}: H M]_{\mathrm{Dv}}$, regardless of the binding mechanism. In conclusion, the hypothesis that PfCRT is a channel for unprotonated CQ species implies that we should not observe any difference in CAR values between CQR and CQS strains, which is in contradiction with the experimentally observed CAR difference (cases B and C in Table 1, main text).

## S2 - Membrane equilibrium equation

The CAR measurements reported by Bray et al [4] are taken after 10 minutes incubation of the cells in the presence of chloroquine. Given the known time evolution of chloroquine uptake [5], the system is expected to have reached a steady state. Consequently, the net chloroquine flux across the membrane of the infected erythrocyte is zero. The only form of chloroquine for which the erythrocyte and the plasmodium membranes are permeable is the un-protonated one and this implies that CQ concentrations are the same on the two sides of the membranes; this concentration is indicated here with $[C Q]_{\mathrm{e}}$. In the case of CQ sensitive (CQS) strains this implies:
$[C Q]_{\mathrm{e}}=[C Q]_{\mathrm{Dv}}$

On the other hand, for $C Q$ resistant ( CQR ) strains, we must take into account the outward chloroquine flux across the vacuole membrane due to PfCRT ( $\mathrm{J}_{\mathrm{PfCRT}}$ ). At steady state the inward diffusive flux of un-protonated chloroquine CQ is balanced by the outward flux due to PfCRT, i.e.

$$
\begin{equation*}
P_{c q}\left([C Q]_{e}-[C Q]_{D V}\right)=J_{P f C R T} \tag{eq-S4}
\end{equation*}
$$

where $\mathrm{P}_{\mathrm{cq}}$ is the permeability of the vacuolar membrane to CQ . Note that equation (eq-S4) becomes equation (eq-S3) when $\mathrm{J}_{\mathrm{PfCRT}}=0$.

## S3 - The expression for the Cellular Accumulation Ratio of chloroquine

We need an expression for the quantity measured in the experiments, i.e. for the Cellular Accumulation Ratio (CAR). Indicating the volume of the infected erythrocyte with $\mathrm{V}_{\mathrm{in}}$, the volume of the digestive vacuole with $\mathrm{V}_{\mathrm{DV}}$ and the concentration of all the chloroquine inside the vacuole with $[\mathrm{C}]_{\mathrm{Dv}}$, we have:

$$
\begin{equation*}
[\mathrm{C}]_{\mathrm{Dv}}=[\mathrm{CQ}]_{\mathrm{Dv}}+\left[\mathrm{CQ}^{+}\right]_{\mathrm{Dv}}+\left[\mathrm{CQ}^{++}\right]_{\mathrm{DV}}+[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{Dv}} \tag{eq-S5}
\end{equation*}
$$

Defining $[\mathrm{C}]_{\mathrm{e}}=\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{e}}$ as the concentration of chloroquine outside the vacuole:

$$
\begin{equation*}
[\mathrm{C}]_{\mathrm{e}}=\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{e}}=[\mathrm{CQ}]_{\mathrm{e}}+\left[\mathrm{CQ}^{+}\right]_{\mathrm{e}}+\left[\mathrm{CQ}^{++}\right]_{\mathrm{e}} \tag{eq-S6}
\end{equation*}
$$

the average chloroquine concentration in the infected erythrocyte is

$$
\begin{equation*}
[C]_{i n}=\frac{[C]_{D V} V_{D V}+[C]_{e}\left(V_{i n}-V_{D V}\right)}{V_{i n}} \tag{eq-S7}
\end{equation*}
$$

The total concentration of chloroquine in each compartment of the infected erythrocyte only depends on the pH of the compartment and on the concentration of the un-protonated chloroquine CQ . Recent measurements $[1,2,3]$ have shown that the pH of the infected erythrocyte cytoplasm and of the plasmodium cytoplasm are very close to physiological pH . Thus, it is reasonable to assume that the pHs of the external medium, of the erythrocyte and of the plasmodium cytoplasm are equal. We indicate
this value as $\mathrm{pH}_{\mathrm{e}}$. Furthermore, due to the permeability of the membrane to CQ , at equilibrium the CQ concentration is equal in all compartments. The CQ concentration and pH being equal inside the erythrocyte and in the external buffer, we have $[\mathrm{C}]_{\mathrm{e}}=$ $[\mathrm{C}]_{\text {out }}$ (as more rigorously shown below in section S4). Therefore, the cellular accumulation ratio CAR can be expressed as:

$$
\begin{equation*}
C A R=\frac{[C]_{\text {in }}}{[C]_{\text {out }}}=\frac{[C]_{D V} V_{D V}+[C]_{e}\left(V_{\text {in }}-V_{D V}\right)}{[C]_{e} V_{\text {in }}} \tag{eq-S8}
\end{equation*}
$$

## S4-Chloroquine concentration is equal in the external buffer and inside the

 erythrocyteLet $\left[\mathrm{CQ}^{\mathrm{TOT}}\right]$ be the concentration of the total amount of free chloroquine, i.e.

$$
\begin{equation*}
\left[\mathrm{CQ}^{\mathrm{TOT}}\right]=[\mathrm{CQ}]+\left[\mathrm{CQ}^{+}\right]+\left[\mathrm{CQ}^{++}\right] \tag{eq-S9}
\end{equation*}
$$

Combining (eq-S9) with eqs. (5) and (6) of the main text, we obtain

$$
\begin{equation*}
\left[C Q^{T O T}\right]=[C Q]\left(1+\frac{\left[H^{+}\right]}{k^{\prime}}+\frac{\left[H^{+}\right]^{2}}{k^{\prime \prime}}\right) \tag{eq-S10}
\end{equation*}
$$

which shows that the total amount of free chloroquine can be expressed as a function of both the pHs and the concentration of the un-protonated form, [CQ]. In our model we make the reasonable assumption that external buffer, erythrocyte cytoplasm, and plasmodium cytoplasm have the same pH . Moreover, at equilibrium, the three compartments have the same [CQ]. Hence, from eq (eq-S10), $\left[\mathrm{CQ}^{\mathrm{TOT}}\right]$ is the same both outside ( $[\mathrm{C}]_{\text {out }}$ ) and inside the infected erythrocyte ( $[\mathrm{C}]_{\mathrm{e}}$ ), with the exception of the vacuole lumen where, due to the acidic pH and to the chloroquine-HM binding, the chloroquine total concentration ([C] $]_{\mathrm{DV}}$, see eq. (eq-S5)) is higher.

S5-CAR expressions in the case of chloroquine-HM binding in linear regime (hypothesis H1)

CAR expressions corresponding to equations (8-10) of the main text are:
$C A R=\frac{V_{e}+V_{D V} \frac{[C Q]_{D V}}{C_{e}}\left(1+(1+\alpha) \frac{\left[H^{+}\right]_{D V}}{k^{\prime}}+\frac{\left[H^{+}\right]_{D V}^{2}}{k^{\prime \prime}}\right)}{V_{e}+V_{D V}}$
$C A R=\frac{V_{e}+V_{D V} \frac{[C Q]_{D V}}{C_{e}}\left(1+\frac{\left[H^{+}\right]_{D V}}{k^{\prime}}+(\alpha+1) \frac{\left[H^{+}\right]_{D V}^{2}}{k^{\prime \prime}}\right)}{V_{e}+V_{D V}}$

$$
C A R=\frac{V_{e}+V_{D V} \frac{[C Q]_{D V}}{C_{e}}(1+\alpha)\left(1+\frac{\left[H^{+}\right]_{D V}}{k^{\prime}}+\frac{\left[H^{+}\right]_{D V}^{2}}{k^{\prime}}\right)}{V_{e}+V_{D V}}
$$

S6 - Proof that $[C Q]_{d v, D}=[C Q]_{e}$ in the hypothesis PfCRT is a channel for the protonated forms of chloroquine

Let $\mathrm{J}_{\text {mem }}$ be the inward membrane flux of unprotonated chloroquine CQ and $\mathrm{J}_{\text {PfCRT }}$ the chloroquine outward flux due to the CQR mutated form of PfCRT. At equilibrium eq (eq-S1) holds, where $\mathrm{J}_{\mathrm{mem}}$ is a function of the chloroquine difference between the plasmodium cytoplasm and the vacuole, i.e. $\mathrm{J}_{\mathrm{mem}}=\mathrm{g}\left([\mathrm{CQ}]_{\mathrm{e}}-[\mathrm{CQ}]_{\mathrm{DV}}\right)$. In particular we have that $\mathrm{J}_{\text {mem }}$ has the following properties:

$$
\begin{align*}
& \mathrm{J}_{\mathrm{mem}}=0 \text { if }[\mathrm{CQ}]_{\mathrm{e}}=[\mathrm{CQ}]_{\mathrm{DV}}  \tag{eq-S11a}\\
& \mathrm{~J}_{\mathrm{mem}}>0 \text { if }[\mathrm{CQ}]_{\mathrm{e}}>[\mathrm{CQ}]_{\mathrm{DV}} \\
& \mathrm{~J}_{\mathrm{mem}}<0 \text { if }[\mathrm{CQ}]_{\mathrm{e}}<[\mathrm{CQ}]_{\mathrm{DV}}
\end{align*}
$$

Let us now consider the case in which $\mathrm{J}_{\text {PfCRT }}$ is a channel for $\mathrm{CQ}^{+}$. In this case we have $\mathrm{J}_{\text {PCCRT }}=\mathrm{f}\left(\left[\mathrm{CQ}^{+}\right]_{\mathrm{Dv}}-\left[\mathrm{CQ}^{+}\right]_{\mathrm{e}}\right)$, and:

$$
\begin{align*}
\mathrm{J}_{\mathrm{PfCRT}}=0 & \text { if } \quad\left[\mathrm{CQ}^{+}\right]_{\mathrm{DV}}=\left[\mathrm{CQ}^{+}\right]_{\mathrm{e}}  \tag{eq-S12a}\\
\mathrm{~J}_{\mathrm{PfCRT}}>0 & \text { if } \quad\left[\mathrm{CQ}^{+}\right]_{\mathrm{DV}}>\left[\mathrm{CQ}^{+}\right]_{\mathrm{e}}  \tag{eq-S12b}\\
\mathrm{~J}_{\mathrm{PfCRT}}<0 & \text { if } \quad\left[\mathrm{CQ}^{+}\right]_{\mathrm{DV}}<\left[\mathrm{CQ}^{+}\right]_{\mathrm{e}} \tag{eq-S12c}
\end{align*}
$$

In experiment $\mathrm{D}, \mathrm{pH}_{\mathrm{e}}=\mathrm{pH}_{\mathrm{DV}}$, hence
$\left(\left[C Q^{+}\right]_{D V}-\left[C Q^{+}\right]_{e}\right)=\frac{\left[H^{+}\right]\left([C Q]_{D V}-[C Q]_{e}\right)}{k^{\prime}}$
(eq-S12a), (eq-S12b) and (eq-S12c) imply
$\mathrm{J}_{\text {PfCRT }}=0$ if $[\mathrm{CQ}]_{\mathrm{DV}}=[\mathrm{CQ}]_{\mathrm{e}}$ (eq-S13a)
$\mathrm{J}_{\text {PfCRT }}>0$ if $[\mathrm{CQ}]_{\mathrm{DV}}>[\mathrm{CQ}]_{\text {e }}$
$\mathrm{J}_{\text {PfCRT }}<0$ if $[\mathrm{CQ}]_{\text {DV }}<[\mathrm{CQ}]_{\text {e }}$
Combining (eq-S11a) with (eq-S13a), it is apparent that $[C Q]_{D V}=[C Q]_{\mathrm{e}}$ is a solution of equation (eq-S1).

In the following we will show that this solution is unique, i.e. that no solution exists in the hypotheses $[\mathrm{CQ}]_{\mathrm{Dv}}>[\mathrm{CQ}]_{\mathrm{e}}$ and $[\mathrm{CQ}]_{\mathrm{Dv}}<[\mathrm{CQ}]_{\mathrm{e}}$.

We first consider [CQ] $]_{\text {Dv }}>[C Q]_{\mathrm{e}}$. In this case, equation (eq-S11c) implies $\mathrm{J}_{\mathrm{mem}}<0$ while equation (eq-S13b) implies $\mathrm{J}_{\text {PfCRT }}>0$, i.e. $\mathrm{J}_{\mathrm{mem}} \neq \mathrm{J}_{\text {PfCRT, }}$, which is not consistent with (eq-S1). Therefore, no solution of eq (eq-S1) exists if we assume [CQ] ${ }_{\text {Dv }}>$ $[\mathrm{CQ}]_{\mathrm{e}}$. A similar argument holds in the hypothesis $[\mathrm{CQ}]_{\mathrm{Dv}}<[\mathrm{CQ}]_{\mathrm{e}}$ : Equation (eqS11b) implies $\mathrm{J}_{\text {mem }}>0$ while equation (eq-S13c) implies $\mathrm{J}_{\text {PfCRT }}<0$, i.e. $\mathrm{J}_{\text {mem }} \neq \mathrm{J}_{\mathrm{PfCRT}}$, that is not consistent with (eq-S1). Therefore no solution of eq (eq-S1) exists if we assume $[C Q]_{D V}<[C Q]_{\mathrm{e}}$.

S7-Value of the channel permeability in the hypothesis PfCRT is a channel and $[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{Dv}}=\alpha[\mathrm{CQ}]_{\mathrm{Dv}}$

The values of un-protonated chloroquine concentration and of PfCRT flux in experiment C have been derived in the main text and are $[\mathrm{CQ}]_{\mathrm{DV}, \mathrm{C}}=1.53 * 10^{-12} \mathrm{M}$ and $\mathrm{J}_{\text {PfCRT }}=2.799 * 10^{-3} \mathrm{nM} \mathrm{cm} / \mathrm{sec}$, respectively. Using these values in the equation of outward PfCRT flux in the case in which PfCRT is a non saturated $\mathrm{CQ}^{+}$channel, i.e. $J_{\text {PfCRT }}=\beta\left(\left[\mathrm{CQ}^{+}\right]_{\mathrm{DV}, \mathrm{C}}-\left[\mathrm{CQ}^{+}\right]_{\mathrm{e}}\right)$, we obtain that the value of the channel permeability $\beta$ is $2.7072 \mathrm{~cm} / \mathrm{s}$. A similar calculation gives $\beta=4.816 * 10^{-3} \mathrm{~cm} / \mathrm{s}$ in the case of PfCRT being a non saturated $\mathrm{CQ}^{++}$channel, i.e. $\mathrm{J}_{\mathrm{PfCRT}}=\beta\left(\left[\mathrm{CQ}^{++}\right]_{\mathrm{DV}, \mathrm{C}}-\right.$ $\left.\left[\mathrm{CQ}^{++}\right]_{\mathrm{e}}\right)$.

S8 - Calculation of $\mathrm{CAR}_{\mathrm{D}}$ in the hypothesis $[\mathrm{CQ}: H M]_{\mathrm{Dv}}=\alpha[\mathrm{CQ}]_{\mathrm{Dv}}(\mathrm{H} 1)$ and

## PfCRT is a non saturated active carrier (J2a) (main text, Figure 1, cells C3-5,5)

## 1) Hypothesis: $J_{P f C R T}=\lambda[C Q]_{D V}$,corresponding to cell $C_{3,5}$ (Figure 1, main text)

The values of un-protonated chloroquine concentration and of PfCRT flux in experiment C have been derived in the main text and are $[\mathrm{CQ}]_{\mathrm{DV}, \mathrm{C}}=1.53 * 10^{-12} \mathrm{M}$ and $\mathrm{J}_{\text {PPCRT }}=2.799 * 10^{-3} \mathrm{nM} \mathrm{cm} / \mathrm{sec}$, respectively. Using these values in the equation of the outward PfCRT flux

$$
\begin{equation*}
\mathrm{J}_{\mathrm{PfCRT}}=\lambda[\mathrm{CQ}]_{\mathrm{DV}, \mathrm{C}} \tag{eq-S14}
\end{equation*}
$$

we obtain $\lambda=18.3 \mathrm{~cm} / \mathrm{s}$. The CQR membrane balance in experiment D (obtained by combining eqs (eq-S4) and (eq-S14)) is:

$$
\begin{equation*}
P_{c q}\left([C Q]_{e}-[C Q]_{D V, D}\right)=\lambda[C Q]_{D V, D} \tag{eq-S15}
\end{equation*}
$$

and can be used to calculate $[\mathrm{CQ}]_{\mathrm{Dv}, \mathrm{D}}$. Substituting $[\mathrm{CQ}]_{\mathrm{DV}, \mathrm{D}}, \boldsymbol{\lambda}$ and $\mathrm{pH}_{\mathrm{DV}}=\mathrm{pH}_{\mathrm{e}}$ in equation (11) of the main text, the $\mathrm{CAR}_{\mathrm{D}}$ value can be easily obtained.
2) Hypothesis: $J_{P f C R T}=\lambda\left[C Q^{+}\right]_{D V}$, corresponding to cell $C_{4,5}$ (Figure 1, main text) In this case the outward PfCRT flux is

$$
\mathrm{J}_{\mathrm{PfCRT}}=\lambda\left[\mathrm{CQ}^{+}\right]_{\mathrm{DV}, \mathrm{C}}
$$

Using $[\mathrm{CQ}]_{\mathrm{Dv}, \mathrm{C}}$ and $\mathrm{J}_{\mathrm{PfCRT}}$ values as derived in the main text, we obtain $\lambda=2.62 \mathrm{x}$ $10^{-4} \mathrm{~m} / \mathrm{s}$. In this case, the CQR membrane balance in experiment D is:

$$
\begin{equation*}
P_{c q}\left([C Q]_{e}-[C Q]_{D V, D}\right)=\lambda\left[C Q^{+}\right]_{D V, D}=\frac{\lambda\left[H^{+}\right]_{D V}[C Q]_{D V, D}}{k^{\prime}} \tag{eq-S16}
\end{equation*}
$$

where eq. (5) (main text) was used. Eq (eq-S16) can be used to calculate [CQ] ${ }_{\text {Dv, }}$. Substituting $[C Q]_{\mathrm{DV}, \mathrm{D}}, \lambda$ and $\mathrm{pH}_{\mathrm{DV}}=\mathrm{pH}_{\mathrm{e}}$ in equation (11) of the main text, the $\mathrm{CAR}_{\mathrm{D}}$ value can be easily obtained.
3) Hypothesis: $J_{P f C R T}=\lambda\left[C Q^{++}\right]_{D V}$, corresponding to cell $C_{5,5}$ (Figure 1, main text)

The outward PfCRT flux is in this case
$\mathrm{J}_{\mathrm{PfCRT}}=\lambda\left(\left[\mathrm{CQ}^{++}\right]_{\mathrm{DV}, \mathrm{C}}\right)$
Using $[C Q]_{D v, C}$ and $J_{\text {PfCRT }}$ values as derived in the main text, we obtain $\lambda=4.81 \times 10^{-7} \mathrm{~cm} / \mathrm{s}$. In this case, the CQR membrane balance in experiment D , is:
$P_{c q}\left([C Q]_{e}-[C Q]_{D V, D}\right)=\lambda\left[C Q^{++}\right]_{D V, D}=\frac{\lambda\left[H^{+}\right]_{D V}^{2}[C Q]_{D V, D}}{k^{\prime \prime}}$
where eq. (6) (main text) was used. Eq (eq-S17) can be used to calculate $[C Q]_{D v, D}$. Substituting $[\mathrm{CQ}]_{\mathrm{DV}, \mathrm{D}}, \lambda$ and $\mathrm{pH}_{\mathrm{DV}}=\mathrm{pH}_{\mathrm{e}}$ in equation (11) of the main text of the main text, the $\mathrm{CAR}_{\mathrm{D}}$ value is obtained.

## S9 - Detailed numerical calculation showing that chloroquine-HM binding in

 experiment $C$ is not in the saturation regionLet us consider the hypothesis that chloroquine-HM binding is in the saturation region in experiment A (cells $\mathrm{C}_{1-5,1-4}$, Figure 1, main text). Combining equation (eq-S5) (eqS6) and (eq-S8), for experiment A (where $\mathrm{pH}_{\mathrm{e}}=\mathrm{pH}_{\mathrm{DV}}$ and $[\mathrm{CQ}]_{\mathrm{e}}=[\mathrm{CQ}]_{\mathrm{Dv}}$ ) the following expression for $[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}, \mathrm{A}}$ is obtained
$[C Q: H M]_{D V, A}=C_{e}\left(C A R_{A}-1\right) \frac{V_{e}+V_{D V}}{V_{D V}}$

If we hypothesized that chloroquine-HM binding in experiment C is in the saturation region, we would have $[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}, \mathrm{A}}=[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}, \mathrm{C}}$ that, combined with (eq-S5) (eq-S6) (eq-S8) and (eq-S18), gives

$$
\begin{equation*}
C A R_{C}=C A R_{A}-1+\frac{C_{e} V_{e}+\left([C Q]_{D V}+\left[C Q^{+}\right]_{D V}+\left[C Q^{++}\right]_{D V}\right) V_{D V}}{C_{e}\left(V_{D V}+V_{e}\right)} \tag{eq-S19}
\end{equation*}
$$

Since the third term of (eq-S19) is always positive, $\mathrm{CAR}_{\mathrm{C}}>\mathrm{CAR}_{\mathrm{A}}-1$, which is not compatible with the experimental data.

S10 - Proof that eq. (18) (main text) is an increasing function of $\left[\mathrm{H}^{+}\right]_{\mathrm{Dv}}$ in the interval $0<\mathbf{p H}_{\text {DV }}<9.15$

In this section we demonstrate that eq. (18) of the main text is an increasing function of $\left[\mathrm{H}^{+}\right]_{\mathrm{DV}}$ in the interval $0<\mathrm{pH}_{\mathrm{DV}}<9.15$ for any value of $\lambda$ and $\mathrm{P}_{\mathrm{cq}}$. For the sake of clarity we report here eq. (18)

$$
\left[C Q^{T O T}\right]_{D V}=[C Q]_{e}\left(1+\frac{\lambda\left[H^{+}\right]_{D V}}{P_{c q} k^{\prime}}\right)^{-1}\left(1+\frac{\left[H^{+}\right]_{D V}}{k^{\prime}}+\frac{\left[H^{+}\right]_{D V}^{2}}{k^{\prime}}\right)
$$

The derivative of $\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{DV}}$ with respect to $\left[\mathrm{H}^{+}\right]_{\mathrm{DV}}$ is

$$
\frac{d\left[C Q^{T O T}\right]_{D V}}{d\left[H^{+}\right]_{D V}}=\frac{\left(\frac{1}{k^{+}}+\frac{2\left[H^{+}\right]_{D V}}{k^{\prime}}\right)\left(1+\frac{\lambda\left[H^{+}\right]_{D V}}{P_{c q} k^{2}}\right)-\frac{\lambda}{P_{c q} k}\left(1+\frac{\left[H^{+}\right]_{D V}}{k^{\prime}}+\frac{\left[H^{+}\right]_{D V}^{2}}{k^{\prime}}\right)}{\left(1+\frac{\lambda\left[H^{+}\right]_{D V}}{P_{c q} k}\right)^{2}}(\text { eq-S20 })
$$

Being the denominator of (eq-S20) always positive, the sign of the derivative will be determined by the sign of its numerator that could be rewritten as

$$
\begin{equation*}
N=\frac{1}{k}+\frac{2\left[H^{+}\right]_{D V}}{k^{\prime \prime}}+\frac{\lambda}{P_{c q} k}\left(\frac{\left[H^{+}\right]_{D V}^{2}}{k^{\prime \prime}}-1\right) \tag{eq-S21}
\end{equation*}
$$

The first two terms are always positive while the third term is positive if $\left[\mathrm{H}^{+}\right]_{\mathrm{DV}}>$ $\left(\mathrm{k}^{\prime \prime}\right)^{-1 / 2}$, i.e. $\mathrm{pH}_{\mathrm{DV}}<9.15$. Therefore, for $\mathrm{pH} \mathrm{DV}<9.15$, the $\left[\mathrm{CQ}^{\text {TOT }}\right]_{\mathrm{DV}}$ expression given in eq. (18) of the main text is an increasing function of $\left[\mathrm{H}^{+}\right]_{\mathrm{DV}}$, regardless the value of $\lambda$ and $\mathrm{P}_{\mathrm{cq}}$.

## S11 - Calculation of CAR $_{D}$ in saturated condition for chloroquine-HM binding and $J_{\text {PfCRT }}=\lambda[C Q]_{\text {Dv }}$ (cells $C_{3,1-4}$, Figure 1 of the main text)

In this case the vacuolar membrane balance equation for $C Q R$ strains reads
$\mathrm{P}_{\mathrm{cq}}\left([\mathrm{CQ}]_{\mathrm{e}}-[\mathrm{CQ}]_{\mathrm{Dv}}\right)=\lambda[\mathrm{CQ}]_{\mathrm{Dv}}$
Eq. (eq-S22) does not depend on the vacuolar $\mathrm{pH}_{\mathrm{Dv}}$ (which alone explains the difference between case C and case D ) hence, being the extra-erythrocytic unprotonated chloroquine concentration ([CQ]e) the same for all experiments, we have $[C Q]_{D V, D}=[C Q]_{D V, C}$. The same argument of section S 14 below holds here, implying $\mathrm{CAR}_{\mathrm{D}}<\mathrm{CAR}_{\mathrm{C}}$, which is not compatible with experimental data. The corresponding cells are shaded in Figure 1 of the main text.

## S12-Calculation of CAR $_{D}$ in saturated condition for chloroquine-HM binding

 and $J_{\text {PfCRT }}=\lambda\left[C Q^{++}\right]_{\text {Dv }}$ (cells C5,1-4,Figure 1 of the main text $]$ )In this case the vacuolar membrane balance equation for CQR parasites reads $P_{c q}\left([C Q]_{e}-[C Q]_{D v}\right)=\lambda\left[\mathrm{CQ}^{++}\right]_{D V}$
that could be rewritten as

$$
\begin{equation*}
[C Q]_{D V}=[C Q]_{e}\left(1+\frac{\lambda\left[H^{+}\right]_{D V}^{2}}{P_{c q} k^{\prime \prime}}\right)^{-1} \tag{eq-S23}
\end{equation*}
$$

Being $\left[\mathrm{CQ}^{\mathrm{TOT}}\right]=[\mathrm{CQ}]+\left[\mathrm{CQ}^{+}\right]+\left[\mathrm{CQ}^{++}\right]$and using the dissociation relations (eqs (5) and (6), main text), we can write

$$
\left[C Q^{T O T}\right]_{D V}=[C Q]_{e}\left(1+\frac{\lambda\left[H^{+}\right]_{D V}}{P_{c q} k^{\prime \prime}}\right)^{-1}\left(1+\frac{\left[H^{+}\right]_{D V}}{k^{\prime}}+\frac{\left[H^{+}\right]_{D V}^{2}}{k^{\prime \prime}}\right)
$$

This expression shows that the free chloroquine inside the vacuole is a function of $\mathrm{pH}_{\mathrm{Dv}}$ and of the ratio between the unknown parameter $\lambda$ and the membrane permeability $\mathrm{P}_{\mathrm{cq}}$. Considering that $\mathrm{pH}_{\mathrm{DV}, \mathrm{C}}$ and $\mathrm{pH}_{\mathrm{DV}, \mathrm{D}}$ are known (in particular $\mathrm{pH} \mathrm{HV}_{\mathrm{DV}, \mathrm{D}}$ $=\mathrm{pH}_{\mathrm{DV}, \mathrm{e}}$ while the $\mathrm{pH}_{\mathrm{DV}, \mathrm{C}}$ value has been derived in the main text), the ratio $\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{DV}, \mathrm{D}} /\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{DV}, \mathrm{C}}$ is a function of $\lambda / \mathrm{P}_{\mathrm{cq}}$ only. The dependence of $\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{DV}, \mathrm{D}} /\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{DV}, \mathrm{C}}$ from $\lambda / \mathrm{P}_{\mathrm{cq}}$ is reported in figure S 1 where it is evident that $\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{DV}, \mathrm{D}} /\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{DV}, \mathrm{C}}$ has an upper bound corresponding to its horizontal asymptote which is equal to 1.198 . Hence we have

$$
\begin{equation*}
\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{DV}, \mathrm{D}}<1.198 *\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{DV}, \mathrm{C}} \tag{eq-S24}
\end{equation*}
$$



Figure S1 - Ratio $\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{DV}, \mathrm{D}} /\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{DV}, \mathrm{C}}$ as a function of $\lambda / \mathrm{P}_{\mathrm{cq}}$. The dashed line corresponds to the horizontal asymptote value $\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{Dv}, \mathrm{D}} /\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{DV}, \mathrm{C}}=1.98$

Let us now analyze separately each hypothesis for the CQ:HM binding:

1) $[C Q: H M]_{D V}=f\left([C Q]_{D V}\right)$, corresponding to cell $C_{5,1}$ (Figure 1 of the main text) Equation (eq-S23) shows that $[\mathrm{CQ}]_{\mathrm{Dv}}$ is a decreasing function of $\left[\mathrm{H}^{+}\right]$, hence, being $\left[\mathrm{H}^{+}\right]_{\mathrm{DV}, \mathrm{D}}<\left[\mathrm{H}^{+}\right]_{\mathrm{DV}, \mathrm{C}}$ we have $[\mathrm{CQ}]_{\mathrm{DV}, \mathrm{D}}>[\mathrm{CQ}]_{\mathrm{DV}, \mathrm{C}}$ and, consequently $[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}, \mathrm{D}}>$ $[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{Dv}, \mathrm{C}}$. This expression, combined with (eq-S24) and eq (15) of the main text does not allow us to draw any conclusion on the relationship between $\mathrm{CAR}_{\mathrm{D}}$ and $\mathrm{CAR}_{\mathrm{C}}$. In particular, (eq-S24) shows that $\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{DV}, \mathrm{D}}<1.198 *\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{DV}, \mathrm{C}}$ while we have demonstrated above that $[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}, \mathrm{D}}>[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{Dv}, \mathrm{C}}$. Hence cell $\mathrm{C}_{5,1}$ cannot be shaded in Figure 1 of the main text.
2) $[C Q: H M]_{D V}=f\left(\left[C Q^{+}\right]_{D V}\right)$, corresponding to cell $C_{5,2}$ (Figure 1 of the main text) Equation (eq-S23) combined with the equilibrium dissociation relation (eq (5), main text) can be rewritten as

$$
\begin{equation*}
\left[C Q^{+}\right]_{D V}=\left(\frac{[C Q]_{e}\left[H^{+}\right]_{D V}}{k^{\prime}}\right)\left(1+\frac{\lambda\left[H^{+}\right]_{D V}^{2}}{P_{c q} k^{\prime \prime}}\right)^{-1} \tag{eq-S25}
\end{equation*}
$$

Equation (eq-S25), together with the known value of $\mathrm{pH}_{\mathrm{Dv}}$, can be used to calculate the ratio $\left[\mathrm{CQ}^{+}\right]_{\mathrm{DV}, \mathrm{D}} /\left[\mathrm{CQ}^{+}\right]_{\mathrm{DV}, \mathrm{C}}$ as a function of $\lambda / \mathrm{P}_{\mathrm{cq}}$ (Figure S 2 ). This ratio is a bounded function ranging between the values 0.0091 and 109.64 , which correspond to the value obtained when $\lambda / \mathrm{P}_{\mathrm{cq}}=0$ and to the horizontal asymptote value, respectively. This implies that no conclusive prediction is possible for [CQ:HM $]_{\mathrm{DV}, \mathrm{D}} /[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}, \mathrm{C}}$ and, therefore, a conclusionon about the relationship between $\mathrm{CAR}_{\mathrm{D}}$ and $\mathrm{CAR}_{\mathrm{C}}$ cannot be drawn. This implies that the current hypothesis cannot be excluded and cell $\mathrm{C}_{5,2}$ cannot be shaded in Figure 1 of the main text.


Figure S2 - Ratio $\left[\mathrm{CQ}^{+}\right]_{\mathrm{DV}, \mathrm{D}} /\left[\mathrm{CQ}^{+}\right]_{\mathrm{Dv}, \mathrm{C}}$ as a function of $\lambda / \mathrm{P}_{\mathrm{cq}}$. The dashed lines correspond to the horizontal asymptote and to the value assumed for $\lambda / \mathrm{P}_{\mathrm{cq}}=0$.
3) $[C Q: H M]_{D V}=f\left(\left[C Q^{++}\right]_{D V}\right)$, corresponding to cell $C_{5,3}$ (Figure 1 of the main text) Equation (eq-S23) combined with equilibrium dissociation relation (eq (6), main text) can be rewritten as

$$
\begin{equation*}
\left[C Q^{++}\right]_{D V}=\left(\frac{[C Q]_{e}\left[H^{+}\right]_{D V}^{2}}{k^{\prime \prime}}\right)\left(1+\frac{\lambda\left[H^{+}\right]_{D V}^{2}}{P_{c q} k^{\prime}}\right)^{-1} \tag{eq-S26}
\end{equation*}
$$

Eq (eq-S26) is an increasing function of $\left[\mathrm{H}^{+}\right]$, which implies that $\left[\mathrm{CQ}^{++}\right]_{\mathrm{DV}, \mathrm{D}}<$ $\left[\mathrm{CQ}^{++}\right]_{\mathrm{DV}, \mathrm{C}}$ and therefore $[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}, \mathrm{D}}<[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{Dv}, \mathrm{C}}$. Let us now demonstrate that
$\mathrm{CAR}_{\mathrm{D}}<1.2 \mathrm{CAR}_{\mathrm{C}}$. Equation (15) (main text) in case D reads
$C A R_{D}=\frac{V_{e}}{V_{e}+V_{D V}}+\frac{V_{D V}\left[C Q^{T O T}\right]_{D V, D}}{C_{e}\left(V_{e}+V_{D V}\right)}+\frac{V_{D V}[C Q: H M]_{D V, D}}{C_{e}\left(V_{e}+V_{D V}\right)}$
that, using eq (eq-S24) and $[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}, \mathrm{D}}<[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}, \mathrm{C}}$, gives

$$
C A R_{D}<\frac{V_{e}}{V_{e}+V_{D V}}+1.2 \frac{V_{D V}\left[C Q^{T O T}\right]_{D V, C}}{C_{e}\left(V_{e}+V_{D V}\right)}+\frac{V_{D V}[C Q: H M]_{D V, C}}{C_{e}\left(V_{e}+V_{D V}\right)}
$$

hence

$$
\begin{equation*}
C A R_{D}<1.2\left(\frac{V_{e}}{V_{e}+V_{D V}}+\frac{V_{D V}\left[C Q^{T O T}\right]_{D V, C}}{C_{e}\left(V_{e}+V_{D V}\right)}+\frac{V_{D V}[C Q: H M]_{D V, C}}{C_{e}\left(V_{e}+V_{D V}\right)}\right) \tag{eq-S27}
\end{equation*}
$$

The expression in brackets on the right side of inequality (eq-S27) corresponds to $\mathrm{CAR}_{\mathrm{C}}$; therefore: $\mathrm{CAR}_{\mathrm{D}}<1.2 \mathrm{CAR}_{\mathrm{C}}$. Since this result is not compatible with the experimental data, we shaded the cell $\mathrm{C}_{5,3}$ in Figure 1 of the main text.
4) $[C Q: H M]_{D V}=f\left(\left[C Q^{\mathrm{TOT}}\right]_{D V}\right)$, corresponding to cell $C_{5,4}$ (Figure 1 of the main text) Equation (eq-S24) implies $[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}, \mathrm{D}}<1.198[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}, \mathrm{C}}$. Following a procedure similar to the one presented in the previous section, it is easy to demonstrate that, in this case, $\mathrm{CAR}_{\mathrm{D}}<1.2 \mathrm{CAR}_{\mathrm{C}}$. Again, being this result not compatible with the experimental data, we shaded cell $\mathrm{C}_{5,4}$ in Figure 1 of the main text.

S13-Calculations relating to each single hypothesis for the chloroquine:HM binding in saturated condition and $J_{\text {PfCRT }}=\lambda\left[C Q^{+}\right]_{\text {Dv }}$ (cells C4,1-4, Figure 1 of the main text])

1) $[C Q: H M]_{D V}=f([C Q])$, corresponding to cell $C_{4,1}$ (Figure 1, main text)

Combining the membrane balance equation (eq (eq-S4)) and the expression for the PfCRT flux $\mathrm{J}_{\text {PfCRT }}=\lambda\left(\left[\mathrm{CQ}^{+}\right]_{\mathrm{DV}}\right)$, we have:

$$
\begin{equation*}
[C Q]_{D V}=[C Q]_{e}\left(1+\frac{\lambda\left[H^{+}\right]_{D V}}{P_{c q} k^{\prime}}\right)^{-1} \tag{eq-S28}
\end{equation*}
$$

(eq-S28) is a decreasing function of $\left[\mathrm{H}^{+}\right]_{\mathrm{Dv}}$. Hence, being $\left[\mathrm{H}^{+}\right]_{\mathrm{DV}, \mathrm{C}}>\left[\mathrm{H}^{+}\right]_{\mathrm{DV}, \mathrm{D}}$, it follows that $[\mathrm{CQ}]_{\mathrm{DV}, \mathrm{C}}<[\mathrm{CQ}]_{\mathrm{DV}, \mathrm{D}}$. The bound chloroquine concentration is an increasing function of [CQ], hence [CQ:HM $]_{\mathrm{DV}, \mathrm{C}}<[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}, \mathrm{D}}$. This result, combined with equation (15) of the main text and with the fact that equation (18) is an increasing function of $\left[\mathrm{H}^{+}\right]_{\mathrm{DV}}$ (see section S 10 ), does not make it possible to predict whether $\mathrm{CAR}_{\mathrm{C}}$ is larger or smaller that $\mathrm{CAR}_{\mathrm{D}}$. Accordingly, the corresponding cell in Figure 1 of the main text cannot be shaded.
2) $[C Q: H M]_{D V}=f\left(\left[C Q^{+}\right]\right)$, corresponding to cell $C_{4,2}$ (Figure 1, main text)

Using the chloroquine dissociation equilibrium (5) (main text), equation (eq-S28) can be rewritten as:
$\left[C Q^{+}\right]_{D V}=\left(\frac{[C Q]_{e}\left[H^{+}\right]_{D V}}{k^{\prime}}\right)\left(1+\frac{\lambda\left[H^{+}\right]_{D V}}{P_{c q} k^{\prime}}\right)^{-1}$
that is an increasing function of $\left[\mathrm{H}^{+}\right]_{\mathrm{Dv}}$. Hence, being $\left[\mathrm{H}^{+}\right]_{\mathrm{DV}, \mathrm{C}}>\left[\mathrm{H}^{+}\right]_{\mathrm{DV}, \mathrm{D}}$, it follows that $\left[\mathrm{CQ}^{+}\right]_{\mathrm{DV}, \mathrm{C}}>\left[\mathrm{CQ}^{+}\right]_{\mathrm{DV}, \mathrm{D}}$. The bound chloroquine concentration, is an increasing function of $\left[\mathrm{CQ}^{+}\right]$hence $[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}, \mathrm{C}}>[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}, \mathrm{D}}$. This result, combined with equation (15) of the main text and with the fact that equation (18) is an increasing function of $\left[\mathrm{H}^{+}\right]_{\mathrm{Dv}}$ (see section S10) implies that $\mathrm{CAR}_{\mathrm{C}}$ is larger than $\mathrm{CAR}_{\mathrm{D}}$, which is not consistent with experimental data. Accordingly, the corresponding cell in Figure 1 of the main text is shaded.
3) $[C Q: H M]_{D V}=f\left(\left[C Q^{++}\right]\right)$, corresponding to cell $C_{4,3}$ (Figure 1, main text)

Using the chloroquine dissociation equilibrium (6) (main text), equation (eq-S28) can be rewritten as:

$$
\left[C Q^{++}\right]_{D V}=\left(\frac{[C Q]_{e}\left[H^{+}\right]_{D V}^{2}}{k^{\prime \prime}}\right)\left(1+\frac{\lambda\left[H^{+}\right]_{D V}}{P_{c q} k^{\prime}}\right)^{-1}
$$

that is an increasing function of $\left[\mathrm{H}^{+}\right]_{\mathrm{Dv}}$. Hence, being $\left[\mathrm{H}^{+}\right]_{\mathrm{DV}, \mathrm{C}}>\left[\mathrm{H}^{+}\right]_{\mathrm{Dv}, \mathrm{D}}$, it follows that $\left[\mathrm{CQ}^{++}\right]_{\mathrm{DV}, \mathrm{C}}>\left[\mathrm{CQ}^{++}\right]_{\mathrm{DV}, \mathrm{D}}$. The bound chloroquine concentration is an increasing function of $\left[\mathrm{CQ}^{++}\right]$, hence $[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}, \mathrm{C}}>[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}, \mathrm{D}}$. This result, combined with equation (15) of the main text and with the fact that equation (18) is an increasing function of $\left[\mathrm{H}^{+}\right]_{\mathrm{DV}}$ (see section S10) implies that $\mathrm{CAR}_{\mathrm{C}}$ is larger than $\mathrm{CAR}_{\mathrm{D}}$, which is
not consistent with experimental data. Accordingly, the corresponding cell in Figure 1 of the main text is shaded.
4) $[C Q: H M]_{\mathrm{DV}}=f\left(\left[C Q^{T O T}\right]\right)$, corresponding to cell $C_{4,4}$ (Figure 1, main text) In section S 10 we showed that $\left[\mathrm{CQ}^{\mathrm{TOT}}\right]$ is an increasing function of $\left[\mathrm{H}^{+}\right]_{\mathrm{DV}}$, hence, being $\left[\mathrm{H}^{+}\right]_{\mathrm{DV}, \mathrm{C}}>\left[\mathrm{H}^{+}\right]_{\mathrm{DV}, \mathrm{D}}$, it follows that $\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{DV}, \mathrm{C}}>\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{DV}, \mathrm{D}}$. The bound chloroquine concentration is an increasing function of $\left[\mathrm{CQ}^{\mathrm{TOT}}\right]$, hence $[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}, \mathrm{C}}$ $>[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}, \mathrm{D}}$. This result, combined with equation (15) of the main text and with the fact that equation (18) is an increasing function of $\left[\mathrm{H}^{+}\right]_{\mathrm{Dv}}$ (see section S 10 ) implies that $\mathrm{CAR}_{\mathrm{C}}$ is larger than $\mathrm{CAR}_{\mathrm{D}}$, which is not consistent with experimental data. Accordingly, the corresponding cell in Figure 1 of the main text is shaded.

## S14-Prediction for $\mathbf{C A R}_{\mathrm{D}}$ in saturated condition for chloroquine-HM binding

$(\mathbf{H} 2)$ and $\mathbf{J}_{\text {PfCRT }}=$ const $(\mathbf{J} 2 b)($ cells C2,1-4,Figure 1, main text $)$
Being $\mathrm{J}_{\text {PfCRT }}=$ const, we have that the vacuolar membrane balance equation for CQR strains reads

$$
\begin{equation*}
\mathrm{P}_{\mathrm{cq}}\left([\mathrm{CQ}]_{\mathrm{e}}-[\mathrm{CQ}]_{\mathrm{DV}}\right)=\text { const } \tag{eq-S29}
\end{equation*}
$$

The extra-erythrocytic unprotonated chloroquine concentration $[\mathrm{CQ}]_{\mathrm{e}}$ is the same for all the experiments hence, from eq. (eq-S29) we have $[C Q]_{D v, D}=[C Q]_{D v, C}$ where the second subscript indicates the experiment ( $\mathrm{A}, \mathrm{B}, \mathrm{C}$ and D in Table 1 of the main text). The pH difference between experiments C and D , namely $\mathrm{pH}_{\mathrm{DV}, \mathrm{D}}>\mathrm{pH}_{\mathrm{DV}, \mathrm{C}}$, implies $\left[\mathrm{CQ}^{+}\right]_{\mathrm{Dv}, \mathrm{D}}<\left[\mathrm{CQ}^{+}\right]_{\mathrm{DV}, \mathrm{C}}$ and $\left[\mathrm{CQ}^{++}\right]_{\mathrm{DV}, \mathrm{D}}<\left[\mathrm{CQ}^{++}\right]_{\mathrm{DV}, \mathrm{C}}$. Being the concentrations of the three free chloroquine forms obtained in case D lower or equal to those obtained in case C, we have $\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{DV}, \mathrm{D}}<\left[\mathrm{CQ}^{\mathrm{TOT}}\right]_{\mathrm{DV}, \mathrm{C}}$, where $\left[\mathrm{CQ}^{\mathrm{TOT}}\right]=[\mathrm{CQ}]+\left[\mathrm{CQ}^{+}\right]+$ $\left[\mathrm{CQ}^{++}\right]$. Moreover, $[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}}$ in case D is equal (if $[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}}=\mathrm{f}([\mathrm{CQ}])$ ) or lower (if $[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}}=\mathrm{f}\left(\left[\mathrm{CQ}^{+}\right]\right)$or $[\mathrm{CQ}: \mathrm{HM}]_{\mathrm{DV}}=\mathrm{f}\left(\left[\mathrm{CQ}^{++}\right]\right)$) than the one obtained in case C. Hence, being both pH and HM -binding contributions, lower in case D than in case $C$, we have $\mathrm{CAR}_{\mathrm{D}}<\mathrm{CAR}_{\mathrm{C}}$ which is not compatible with experimental data. The corresponding cells are shaded in Figure 1 of the main text.

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[^0]:    * These authors contributed equally

