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RESEARCH ARTICLE

# Comparing the salinity tolerance of twenty different wheat genotypes on the basis of their physiological and biochemical parameters under NaCl stress

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# Abstract

The climate has drastically changed over the past two decades. Rising temperatures and climate change may lead to increased evapotranspiration, specifically soil evaporation, causing water to evaporate and salt to accumulate in the soil, resulting in increased soil salinity. As a result, there is a need to evaluate methods for predicting and monitoring the effects of salinity on crop growth and production through rapid screening. Our study was conducted on 20 wheat genotypes, 10 sensitive and 10 tolerant, exposed to two salinity levels (90 and 120 mM NaCl) with the control under greenhouse conditions. Our results revealed significant differences in the genotypes' response to salinity. Salt stress decreased chlorophyll index in sensitive genotypes but increased chlorophyll *a* and carotenoids in tolerant genotypes at 90 mM. Salt stress also increased protein, proline, lipoxygenase, and reactive thiobarbituric acid levels in all wheat genotypes. The study suggests that plant photosynthetic efficiency is a reliable, non-destructive biomarker for determining the salt tolerance of wheat genotypes, while other biochemical traits are destructive and time-consuming and therefore not suitable for rapid screening.

# 1. Introduction

Wheat (*Triticum aestivum* L.) is an important staple food, providing nearly 30% of the world's population (4.5 billion people) with calories and 20% of total protein requirements [1]. It is grown in many countries around the world to meet the food needs of the population.

However, wheat yield per hectare is much lower than its capacity for several reasons, the most common being salinity, as shown by *Pramila*, *et al.* [2].

Salinity stress, which is caused by high salt levels in the soil, can negatively impact plant growth and development. It can lead to reduced water uptake, nutrient imbalances, and damage to the plant's cells and tissues [3]. The detrimental effect of salinity stress basically occurs in two successive phases: (i) osmotic stress, and (ii) ionic toxicity (Na<sup>+</sup> and Cl<sup>-</sup>), followed by the subsequent impact of secondary stresses such as oxidative stress and nutritional imbalances [4,5]. Low or moderate salinity mainly causes osmotic stress; it also affects the photosynthetic activity of the plant, resulting in low growth and yield [6]. It also weakens the light-collecting pigments known as photosynthetic pigments, which are present in the thylakoid membranes of chloroplasts and include carotenoids, chlorophyll *a*, and chlorophyll *b*. High soil salinity lowers wheat leaf water potential, turgor pressure, and stomata closure, as well as CO<sub>2</sub> conductance across the stomata, cell wall integrity, oxidative stress, and toxic metabolite synthesis, all of which lead to ultimate plant death [2]. The stomatal pattern of gas conductance is reduced by salt stress, resulting in inadequate CO<sub>2</sub> supply to wheat plants, which contributes to increased production of reactive oxygen species (ROS) [6].

Osmotic adjustment is a plant's ability to maintain turgor pressure and water content in the cells despite changes in water potential. This is achieved through the accumulation of solutes, such as sugars and amino acids, in the cells, which increases the osmotic potential and helps to maintain water uptake and turgor pressure. This can help plants to survive in saline soils where the water potential is low, and it allows the plant to maintain a better growth and development [7]. In other words, osmotic and ion toxicity effects were thought to be spatially and temporally separated. This spatial and temporal separation suggested that early salinity stress responses are due to general osmotic or water deficit stress and that sodium-specific responses (i.e., ion sequestration or exclusion) are induced later [8]. Salinity adaptation mechanisms such as osmotic adjustment and cellular exclusion and compartmentalization of Na<sup>+</sup> ions play a role in alleviating the deleterious effects of salinity [9].

To protect and preserve osmotic stability, cells accumulate proline, which is probably the most widely distributed osmolyte found in plants and other organisms [10]. Proline is a compound that tends to accumulate in response to metabolic salt stress [11]; it is thus important in the osmotic adjustment in plants under stress and serves as an osmoprotectant [12]. A high level of proline in the cytosol reduces the cellular water potential below the external water potential, enhancing the water flow into the cells to maintain cellular water status and plant cell turgidity [13]. Apart from acting as an osmolyte for osmotic adjustment, proline contributes to stabilizing subcellular structures (e.g., proteins and membranes) [14], buffering cellular redox potential against stresses [15,16].

Photosynthesis is the major source of energy that has significant implications for all aspects of plant metabolism and physiology [17]. The redox status of plant cells is mainly determined by photosynthesis, which is why they are at the center of regulatory networks [18]. As a result, the analysis of plant photosynthetic efficiency based on measurements of chlorophyll fluorescence parameters is recommended as an accurate tool for evaluating plant responses to unfavourable photosynthesis environmental conditions and their impact on the plants [19]. Accordingly, the evaluation of the central role of photosynthesis in plant phonotyping is very important [20,21].

Chlorophyll-*a* fluorescence has been described as a re-emission of absorbed light that plants cannot use in the photochemical process of photosynthesis. The inverse relationship between fluorescence kinetics and photosynthesis helps us to understand the biophysical processes of photosynthesis. Measurement of chlorophyll-*a* fluorescence is a valuable non-invasive tool that has been used in eco-physiological studies and extensively used to evaluate the response of plants to environmental stress [22].

The JIP-test is a quick and efficient way to assess the effectiveness of acclimation procedures. It has been used to study the relationship between light-dependent processes and chlorophyll-*a* fluorescence and also when selecting crops for salt and drought tolerance [23–25]. The concept of "energy flux" through thylakoid membranes serves as the basis [26]. The equilibrium between the total energy inflows and outflows for each of the light-collecting complexes studied can be represented by the operationalized simple algebraic equations of this theory, which also reveal the probable distribution of absorbed energy. These equations can be used to explain how the photosystem II (PSII) complexes interact energetically (also known as "grouping", "connectivity", and "overall grouping probability") [14].

The primary goal of this study was to assess the effects of salt stress on 20 wheat genotypes using chemical and physiological properties, including photosynthetic efficiency of plants using chlorophyll-*a* fluorescence measurements. This may help to understand the mechanisms behind the variations in their tolerance and to test which parameters can be used as biomarkers for rapid monitoring of plants growing under salt stress.

# 2. Materials and methods

#### 2.1. Plant material and salt treatments

In this study, 20 genotypes were selected from a population of 240 native (Iranian) wheat landraces that had been previously tested for stress resistance by the Agricultural Research Institute. These 20 genotypes were chosen as the best genotypes from the previous study (Table 1). The surface of the seeds was sterilised in 3% H<sub>2</sub>O<sub>2</sub> for 20 minutes and then ten times in distilled water. In 2018, the experiment was conducted at the Zabol University Research Greenhouse using a randomized complete block design (RCBD) with two replicates. The treatments included two salinity levels of 90 and 120 mM NaCl, with three gradual steps-initiation, and a control treatment. Seedlings were hydroponically cultivated in custom- made plastic trays

|                     | Ent | Cid    | GID     | Taxon                             | Origcty | Collsite   |
|---------------------|-----|--------|---------|-----------------------------------|---------|------------|
| Sensitive genotypes | 77  | 176907 | 189040  | Triticum aestivum subsp. aestivum | IRAN    |            |
|                     | 81  | 176948 | 189193  | Triticum aestivum subsp. aestivum | IRAN    | -          |
|                     | 101 | 350316 | 189956  | Triticum aestivum subsp. aestivum | IRAN    | Hamedan    |
|                     | 120 | 268851 | 283138  | Triticum aestivum subsp. aestivum | IRAN    | Zanjan     |
|                     | 124 | 348935 | 283449  | Triticum aestivum subsp. aestivum | IRAN    | Mashhad    |
|                     | 126 | 348960 | 283553  | Triticum aestivum subsp. aestivum | IRAN    | Mashhad    |
|                     | 127 | 349005 | 283602  | Triticum aestivum subsp. aestivum | IRAN    | Mashhad    |
|                     | 204 | 267912 | 375626  | Triticum aestivum subsp. aestivum | IRAN    | Kerman     |
|                     | 210 | 267763 | 375620  | Triticum aestivum subsp. aestivum | IRAN    | Kerman     |
|                     | 213 | 268362 | 375743  | Triticum aestivum subsp. aestivum | IRAN    | Kerman     |
| Tolerant genotypes  | 2   | 299978 | 319956  | -                                 | -       | -          |
|                     | 11  | 178271 | 187505  | Undetermined sp.                  | IRAN    | Saghez     |
|                     | 86  | 176978 | 189280  | Triticum aestivum subsp. aestivum | IRAN    | -          |
|                     | 109 | 177264 | 190095  | Triticum aestivum subsp. aestivum | IRAN    | Kermanshah |
|                     | 151 | 350238 | 374133  | Undetermined sp.                  | IRAN    | Ilam       |
|                     | 191 | 349936 | 375454  | Triticum aestivum subsp. aestivum | IRAN    | Mashhad    |
|                     | 199 | 350063 | 375564  | Triticum aestivum subsp. aestivum | IRAN    | Mashhad    |
|                     | 205 | 268045 | 375659  | Triticum aestivum subsp. aestivum | IRAN    | Kerman     |
|                     | 232 | 179295 | 375963  | Triticum aestivum subsp. aestivum | IRAN    | Esfahan    |
|                     | 239 | 350576 | 2437249 | Triticum aestivum subsp. aestivum | IRAN    | Tehran     |

Table 1. Information on sensitive and tolerant genotypes.

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(Width 60 cm, length 60 cm, and height 20 cm; Luoxi Plastic Products Co., Shandong, China). Ten seeds were placed in each hole before the trays were placed in other rectangular plastic trays with 4.0 L of Hoagland's solution [27]. These solutions were continuously aerated by electric pumps and renewed every 7 days. Salt treatments were initiated after germination and lasted about 3 weeks. HCl or KOH were used to maintain the pH of the solution at 6.5 throughout the experiment. The study was performed in growth chambers under artificial light (fluorescent lamp photosynthetic photon flux density (PPFD) of 150  $\mu$ mol m<sup>-2</sup> s<sup>-1</sup>). NaCl solutions with concentrations of 90 mM and 120 mM were used to feed the corresponding treatments to test the effects of different salinity levels on 20 wheat genotypes. The pH of Hoagland's nutrient solutions and EC were adjusted to 6–6.5 and 20 mM with HCl or KOH in the control treatment, respectively. Five samples were used for each replication. Samples were stored at -80°C until chemical parameters were measured.

### 2.2. Determination of chlorophyll a fluorescence

Chlorophyll fluorescence signals were measured using HandyPEA portable fluorometer (Hansatech Instruments Ltd., Norfolk PE32IJL, England) (Table 2). First, plants were adapted in the dark for at least 30 min using leaf clips (4 mm diameter cross section) and a red LED saturation pulse of 3500  $\mu$ mol photon m<sup>-2</sup> s<sup>-1</sup>. Measurements were made in the middle of fully developed leaves of both the control and salt-stressed plants. For each treatment, two measurements were taken on five different plants (10 repetitions as a total for each treatment).

## 2.3. Determination of photosynthetic pigments

After 21 days, pigment contents were analysed and quantified using techniques proposed by *Harborne* [30]. Fresh leaf samples were washed with deionized water to remove impurities from the surface. Then, 1 g. of the leaf tissue was used to extract pigments with 80% acetone. According to *Knight and Mitchell* [31], the absorbance values for Chl *a*, Chl *b*, and Car. were

| JIP | Interpretation |
|-----|----------------|

Table 2. Definition of selected JIP test parameters [28,29].

| parameter           | interpretation   |
|---------------------|--|
| Fo                  | Minimal fluorescence   |
| F <sub>M</sub>      | Maximal fluorescence   |
| Фро                 | Maximal quantum yield of the primary photochemical reaction in PSII in dark-adapted samples  |
| $\Phi_{E0}$         | Quantum yield of the process of electron transfer from $Q_A$ -to electron carriers beyond $Q_A$ -  |
| $\psi_{E0}$         | The probability that the energy of an exciton trapped by active PSII reaction centre (RC) will be utilized for electron transport beyond $\rm Q_A$                             |
| Sm                  | otal electron carriers, per RC, reduced during the time of the induction rise (from $F_0$ to $F_M$ )   |
| Ν                   | urnover number, expressing how many times $Q_A$ is reduced until $F_M$ is reached  |
| M <sub>0</sub>      | Approximated initial slope (in $ms^{-1}$ ) of the fluorescence transient normalized to the maximal variable fluorescence. Reflects the maximal rate of initial $Q_A$ reduction |
| RE <sub>0</sub> /RC | Flux of electrons reaching the end carriers at the acceptor side of PSI as per RC  |
| ABS/RC              | Absorption flux (exciting PSII antenna Chl <i>a</i> molecules) per RC  |
| RC/CS <sub>0</sub>  | Number of active (QA reducing) PSII reaction centres per cross section   |
| δRo                 | Probability with which an electron from the intersystem electron carriers is transferred to reduce end electron acceptors at the PSI acceptor side                             |
| PI <sub>abs</sub>   | Performance index for energy conservation from photons absorbed by PSII until the reduction of intersystem electron acceptors  |
| PI <sub>total</sub> | Performance index for energy conservation from photons absorbed by PSII until the reduction of PSI end electron acceptors  |

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assessed using a UV-visible spectrophotometer (Model BTS-45, United Kingdom) at three wavelengths, 665, 649, and 470 nm, respectively. And the following formulas were used to determine the outcomes: [Chl *a* (mg g<sup>-1</sup> FW) = (13.95OD<sub>665</sub>–6.88OD<sub>649</sub>)V/200 W]; [Chl *b* (mg g<sup>-1</sup> FW) = (24.96OD<sub>649</sub>–7.32OD<sub>665</sub>)V/200W]; [Car. (mg g<sup>-1</sup> FW) = (1000OD<sub>470</sub>–2.05Chl a–114.80Chl b) V/ (245×200 W)]

Where; Chl *a*-chlorophyll *a*, Chl *b*-chlorophyll *b*, Car.-carotenoid, V-volume, and W-sample weight, FW-Fresh weight.

## 2.4 Determination of Na<sup>+</sup> and K<sup>+</sup>

The content of sodium and potassium components in the leaves of wheat genotypes was determined after 21 days of salt stress under hydroponic conditions. They were placed in the oven for drying. The dry weight of the leaves was determined. The dried leaves were placed in Falcon tubes containing 25 mL of a 1% HNO<sub>3</sub> solution. A solution with a volume of 10 mL was prepared to test K<sup>+</sup> and Na<sup>+</sup>. Using a flame photometer (JENWAY, model: PFP7, U.K.), leaf samples were subjected to *Shavrukov*, *et al.* [32].

#### 2.5. Measurement of Thiobarbituric Acid Reactive Material (TBARM)

The amount of malondialdehyde, the end product and generally stable byproduct of the oxidation process of large molecules, is used to calculate the TBARM value, which is a measure of oxidative stress. In this case, a modified version of the approach of *Harborne* [30] was used. 1 mL of chloroacetic acid (15% w/v) was applied to 0.5 g of homogenised leaves. After the addition of 10 mL of acetone, the mixture was mixed vigorously before centrifugation at 4750 rpm for 15 min. The small precipitate produced by centrifugation was washed with five millilitres of acetone. After vortexing, centrifugation was again performed at the same speed for 10 minutes, repeating the last step four times. The solution was then heated to 100°C for 30 minutes with the addition of 3 millilitres of phosphoric acid (1 wt%) and 1 millilitre of thiobaric acid (0.6 wt%). The reaction was stopped by rapidly cooling the tubes in ice, and the resulting solution's adsorption amount was measured at 532 and 590 nm using an optical adsorption apparatus (BR Technologies model BT 600). Last but not least, the amount of TBARM in 1 g. fresh weight was measured as Fresh Weight (FW).

#### 2.6. Measurement of Lipoxygenase (LOX)

To measure the amount of LOX, 2.5 g of the leaf sample was placed in cold water and then centrifuged (12000 g for 10 minutes). After centrifugation, the top solution was removed, and the rest was purified using the PD-10 gel column. An equal amount of potassium phosphate buffer (pH = 6.6 mm) was added, and sodium and linoleic acid buffer (80 nmol) were added to the homogeneous solution and measured at 234 nm using a spectrophotometer (model BTS-45, United Kingdom) [33].

#### 2.7. Measurement of proteins amount

Protein content was measured by *Bradford* [34] method using a spectrophotometer (model BTS-45, United Kingdom) at 595 nm. This method is based on the binding of Comaxi Briant Blue G250 in an acidic reagent to a protein molecule.

#### 2.8. Measurement of proline levels

Measurement of proline was performed by the method of *Bates*, *et al.* [35]. 0.1 g. of freshly beaten leaf tissue was poured into 15 mL of Falcons with liquid nitrogen and 10 mL of

sulfosalicylic acid 3% was added to the samples. The solvent was centrifuged at 13,000 rpm for 10 minutes. After centrifugation, 2 milliliters of ninhydrin solution were added to the samples. The solution was placed in a steam bath at 100°C for 1 hour. Then, the tubes were placed in a mixture of water and ice for 10 minutes to prevent a further reaction. After heating with the medium, 4 mL of toluene was added to the samples. After vortexing, the absorbance was read at 520 nm using a spectrophotometer (model BTS-45, United Kingdom).

#### 2.9. Statistical analysis

The statistix 8.1 programme was used to examine all data using the variance test (One way-ANOVA) [36,37]. With a p-value of  $p \le 0.05$ , Fisher's least significant difference test was used to compare the means of each characteristic within each genotype.

# 3. Results and discussion

#### 3.1. Chlorophyll a fluorescence

Twenty wheat genotypes (ten sensitive genotypes and ten tolerant genotypes) were selected for salinity tolerance test from the 240 wheat genotypes (Fig 1) using 2 parameters: Total Performance Index ( $PI_{total}$ ) and Performance Index of Absorbance ( $PI_{abs}$ ). Chlorophyll fluorescence induction curves (logarithmic time scale of 0.01 to 1.000 ms) were generated to evaluate the differences between stress and control plants after twenty-one days of stress application (Figs 2 and 3).

To present the obvious figures, the graph was divided into two parts (tolerant and sensitive genotypes), each containing ten graphs of wheat genotypes (Figs 2–5). The effect of different salinity on the transient fluorescence curve of the sensitive genotypes of wheat landraces was very significant after 21 days compared to the tolerant genotypes (Figs 2 and 3). The flat fluorescence curve of the sensitive genotypes revealed that the response of photosynthesis process to salinity stress was quite rapid, and strong (Fig 2). Salinity had a significant negative effect on the 0 and P stages of the curve in the sensitive genotypes (Fig 2). In contrast, salinity showed no significant effect on these bands in the tolerant genotypes (Fig 3).



Fig 1. Twenty wheat genotypes (ten sensitive and ten tolerant genotypes) were grown in hydroponic boxes exposed to various concentrations of NaCl.

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Fig 2. Chlorophyll *a* fluorescence induction curve of 10 sensitive wheat genotypes exposed to various NaCl concentrations after 21 days.

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Applied salt stress resulted in a significant increase in O-J phase in some sensitive genotypes (G77, G124, G126, G127, G210, G210, and G213) as compared to tolerant genotypes (Figs 2 and 3). Under conditions of 120 mM, the J-P phase of the chlorophyll fluorescence transient

-120 mM

90 mM

Control



Fig 3. Chlorophyll *a* fluorescence induction curve of 10 tolerant wheat genotypes exposed to various NaCl concentrations after 21 days.

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curves decreased significantly in sensitive genotypes. Additionally, the typical appearance of J-I phase of the OJIP curve was completely lost in all sensitive genotypes. The shape of the chlorophyll fluorescence transient curve was typical in the tolerant genotypes under the same

conditions (Fig 3). Moreover, under 90 mM NaCl, the typical J-I phase of the OJIP curve was completely lost in several sensitive genotypes (G120, G124, G204, and G213). Under salt stress, the J-P phase of the chlorophyll fluorescence transient curve was significantly decreased in the sensitive genotypes.

The experiment was conducted not only to evaluate the response of different genotypes to salt stress but also to explore the significance of a rapid fluorescence approach for phenotyping of wheat. Previous studies have shown that physiological parameters such as  $PI_{abs}$  and  $PI_{total}$ , which are directly related to the function of the photosynthetic apparatus, can be used as reliable indicators or biomarkers of salt tolerance in wheat [38].

The spider plot revealed that under salt stress, there was a decrease in  $F_M$ ,  $F_V$ ,  $F_0$ ,  $ET_0/RCsm$ ,  $PI_{total}$ , and  $PI_{abs}$  as indicated by the parameter's values (Figs 4 and 5). Compared with the control, an increase in  $F_0$ ,  $V_J$ ,  $F_V/F_0$ ,  $DI_0/RC$ , and ABS /RC values was observed. No significant differences were observed between the control and salt-tolerant genotypes in fluorescence indices (fluorescence at time point  $-F_0$ , maximum fluorescence intensity  $-F_M$ ), and ratio between photochemical and non-photochemical quantum efficiency  $-F_V/F_0$ ), electron transport quantum yield ( $\varphi_{E0}$ ), vitality index (power index on an absorbance basis), and specific energy fluxes ( $ET_0/RC$  and  $DI_0/RC$ ) (Fig 4). Significant differences were detected in sensitive genotypes under salt stress, but a significant decrease in  $F_V/F_0$ ,  $ET_0/RC$ ,  $F_M$ ,  $F_V, F_V/F_0$ ,  $ET_0/RC$ , Sm, and absorption-based performance ( $PI_{abs}$ )index and  $PI_{total}$ , and an increase in  $V_J$ , ABS /RC, and  $DI_0/RC$  (dissipation energy flux genotypes, the absorption flux per reaction center (ABS /RC), dissipation energy flux per RC ( $DI_0/RC$ ), minimum fluorescence ( $F_0$ ), number of  $Q_A$  redox conversions to  $F_M$  (N), size of the pool of electron acceptors on the reducing side of PSII (Sm), and loss of energy absorbed in antennas ( $\phi$ Do) were found to vary significantly between 90 mM and 120 mM (Fig 5).

The electron transport chain and chloroplast-based carotenoid and chlorophyll production are two important environmental targets [39]. While phosphorylation and NADP photoreduction occur via the electron transport chain, its electron carriers, and enzymes, chlorophyll and carotenoid production can be linked to the Light Harvest Complexes (LHC) and photosynthetic reaction centres antennae [40]. The JIP assay and its parameters can be used to identify and evaluate the change at these two targets. The light is re-emitted with the help of chlorophyll molecules when returning from excited to unexcited state, which is called chlorophyll fluorescence. It is used as an index of photosynthetic energy conversion in higher plants, bacteria and algae [41]. Excited chlorophyll, which drives photosynthesis, emits the assimilated light energy as heat in non-photochemical suppression or by mission as fluorescence radiation [42]. Because these services are complementary, the study of chlorophyll fluorescence is an important tool in plant research with a wide range of applications [43]. When the total amount of  $Q_A$ is oxidised, the initial chlorophyll fluorescence at the O band shows the lowest fluorescence yield [44]. The P band corresponds to the state of fully assembled Q<sub>A</sub> molecules in the moderated state. The J and I bands are measured at 2 and 30 milliseconds, respectively. The transition from stage O to J is brought about by the conversion of  $Q_A$  to  $Q_A^-$  and is associated with the main photochemical processes of PSII. Both the presence of fast and slow reducing PQ centres and the different redox states of PSII reaction centres are reflected in the intermediate step I and final step P. The OJIP transient represents the gradual reduction of the electron transport pool of PSII [45].

The intensity of fluorescence in the OJIP transient curve decreases with increasing NaCl loading concentration, as shown in Figs 2 and 3. The typical polyphasic transient was sensitive to 120 mM salinity treatment in all genotypes. The J-I and I-P phases were decreased under these conditions (Fig 2). While in the tolerant genotypes under the same conditions, the shape of the chlorophyll-*a* fluorescence curve was typical (Fig 3). Also, the O-J phase of the



Fig 4. Spider plot presenting the JIP-test parameters (Table 2) calculated from 10 sensitive genotypes of wheat exposed to various concentrations of NaCl after 21 days.

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chlorophyll-a fluorescence transient curve in sensitive genotypes under salt stress decreases sharply under 90 mM conditions, but it drastically decreased in some sensitive genotypes, including G120, G124, G204, and G213 (Fig 2). The J-I phase of the curve corresponds to the decrease in the secondary electron acceptors  $Q_B$ , plastoquinone (PQ), cytochrome  $b_6 f$  (Cyt  $b_6$ , and plastocyanin (PC). Also, the typical multiphase transient was saline when treated with 120 mM, and the O-J phase was significantly decreaed under these conditions in all sensitive genotypes except G77, G120, and G126. Different sections of the transient fluorescence curve or OJIP curve show different photosystem events. The I-P phase reflects the rate of ferredoxin reduction and is used as a measure of the relative abundance of PSI compared to PSII, while the J-I phase is responsible for the chlorophyll fluorescence quenching that characterises the activity of the water diffusion complex on the PSII donor side [45,46]. The O-J phase is characterised by the gradual decrease of QA, the main electron acceptor in PSII. When this is not the case, it is related to the proportional size of the final PSI electron acceptor pools [47,48]. The reduction of electron transport on the donor side of PSII to the reaction centres for electron transfer from the water decomposition system to PSI or the shrinkage of the pool size of electron acceptors in PSII ( $Q_A$ ,  $Q_B$  and PQ pools) could be the cause of a decrease in fluorescence yield in phases J, I, and P [41,49].

From the data calculated by the JIP test, shown in Figs 4 and 5, it is evident that for all tolerant genotypes,  $F_V/F_M$ , the maximum quantum yield of PSII photochemistry,  $F_V/F_0$ , and the activity of the water-splitting complex on the donor side of PSII were not significantly changed by salinity after 21 days (Fig 5). These studies indicate that salt stress has no effect on electron transfer rates on the donor side of PSII. However,  $F_V/F_M$  and  $F_V/F_0$  changed significantly in all sensitive genotypes after salinity (Fig 4). It can be said that salt stress significantly negatively affects electron transfer rates on the donor side of PSII [22]. Increasing salinity in the sensitive genotypes showed a significant increase in  $F_0$ . However, photosynthetic apparatus was not significantly changed in the tolerant genotypes, whereas  $F_M$  significantly decreased in the sensitive genotypes.

There may be a number of reasons for the increased  $F_0$  levels observed in susceptible genotypes under salt stress. One of them could be an increase in the number of inactive reaction centres in which electrons cannot be transported from the decreasing  $Q_A$ , resulting in higher measured  $F_0$ . The separation of LHC II from the PSII core may have been the cause of poor energy transfer from LHC II to the PSII reaction centre, leading to stronger fluorescence from LHCII [50].

The  $F_M$  increase reflects the decreased number of nonreducing  $Q_A$  reaction centers. The  $F_0$  increase and decrease in maximum fluorescence intensity indicate blockage of electron transport to  $Q_A$  and result in significantly lower radiation dissipation of the excited states of the photosystem II antennae chlorophylls [51]. The decrease in  $F_V/F_M$  may be related to a decrease in photosystem II activity and/or a decrease in photochemical activity, indicating a problem with the functioning of the photosynthetic apparatus [52]. Our results showed that salinity strongly increases the maximum electron transport flux per (ET\_0/ RC), the absorption flux per reaction center (ABS/RC), the dissipation energy flux per (DI\_0/ RC), and the relative variable fluorescence at the J-step (V\_J). Increased salinity magnifies significant differences in chlorophyll-*a* fluorescence parameters such as the quantum yield for reduction of the end electron acceptor on the acceptor side of photosystem I ( $\varphi_{R0}$ ), the quantum yield of electron transport beyond  $Q_A$  ( $\varphi_{E0}$ ), and the probability (at time 0) that a trapped exciton will shift an electron into the electron transport chain beyond  $Q_A$  ( $\psi_{E0}$ ) in sensitive genotypes.

The total amount of photons captured by chlorophyll molecules in all RCs is multiplied by the total number of active RCs to obtain the effective antenna size of active RCs. The value of

ABS /RC increased with increasing salt concentration, which decreased the antenna size of active RCs (Figs 4 and 5).

The kinetics of relative (V<sub>J</sub>) were determined using the formula  $[V_J = F_J - (F_0/F_M) - F_0]$  to localise the effect of salt stress in the electron transport chain on the acceptor side of the photosystem II. Relative (V<sub>J</sub>) for photosystem II -units shut down = the fraction of RCs closed in the J stage, expressed as a percentage of the total number of RCs that can be closed [53], F<sub>J</sub> is the fluorescence at the J stage. Figs 4 and 5 show the results consistent with *Kalaji*, *et al.* [54], *Kalaji*, *et al.* [22], and *Kalaji*, *et al.* [55].

After salinization, the  $\psi_{E0}$  values increased significantly below the values of the stressed samples. Thus, it can be said that the salt stress on the  $\Psi_{E0}$  values of the salt-stressed samples was high and the electron supply from the PSII donor side (OEC) was low because the carriers efficiently transferred the electrons to the next step of the electron transport chain. This indicates that salinity had a greater effect on the reduction side of PSII than on the  $\Psi_{F0}$ -acceptor side, where  $_{\rm F0}$ : denotes the reduction of the acceptor side of PSII [41]. Indeed, the parameter  $\Psi_{E0}$  is associated with the balance between the efficiency and inefficiency of dark responses after QA treatment, which was significantly increased by 120 mM treatment. This result suggests that salinity induces the redox reaction after  $Q_A$  due to the decreased connectivity of electron flow from  $Q_A$  to  $Q_B$  [56]. In certain genotypes, the  $\varphi_{R0}$  parameter has a similar trend to the  $\psi_{E0}$  value that characterises the PSII acceptor site. It appears that PSI accepts almost all electrons when the electron supply on the donor side is limited [56]. At increased salinity, the  $\varphi_{E0}$  value decreases very sharply and also shows the highest quantum yield for electron transfer via  $Q_A$  [57]. Thus, higher  $\varphi_{E0}$  is important for the effectiveness of photosynthetic electron movement in susceptible genotypes of wheat under salt stress (Fig 5). It appears that a greater amount of energy was used to restore QA in the plants.

After 21 days of salt stress, the parameters sm, N, performance index on an absorption basis ( $PI_{abs}$ ), and performance index ( $PI_{total}$ ) affected salinity significantly more in the sensitive genotypes than in the tolerant ones. Under salt stress, there were very significant decreases in both  $PI_{abs}$  and  $PI_{total}$  parameters.  $PI_{abs}$  is also a parameter that reflects the performance and condition of the photosynthetic apparatus, except that it contains three parameters (RC/ABS,  $TR_0/DI_0$ , and  $ET_0/(TR_0-ET_0)$  [58]. Since these three elements are interrelated, the performance index can more accurately reflect the state of the photosynthetic apparatus [59].  $PI_{abs}$  is extremely sensitive than  $\varphi_{P0}$  and can reflect the effects of stress on the photosynthetic apparatus much better [60]. The  $PI_{abs}$  and  $PI_{total}$  performance indices were used in this work to evaluate performance to loss of PSI end electron acceptors and to quantify PSII performance. In addition, we found that  $PI_{abs}$  and  $PI_{total}$  are more sensitive than  $F_V/F_M$ , because although the trend of  $PI_{abs}$  is similar to that of  $F_V/F_M$ , the degree of change was much greater for  $PI_{abs}$  than for  $F_V/F_M$  (Figs 4 and 5). Therefore, according to this study,  $PI_{abs}$  and  $PI_{total}$  may be better parameters to study the change of PSII and measure the power to reduce PSI end electrons during stress than ( $F_V/F_M$ ).

The result of our study based on two parameters, performance index for absorption ( $PI_{abs}$ ) and  $PI_{total}$ , indicated that the tolerant genotypes of wheat were much more resistant to salt stress than the sensitive genotypes. Accordingly, these photosynthetic traits could be good indicators of wheat adaptation to salinity. Consequently, the measurements for these traits are nonintrusive, rapid, and credible; the approach is quite remarkable.

The activities of the tolerant genotypes showed increased Na<sup>+</sup> and Na/K levels under salt conditions, but the increase was higher in the sensitive genotypes. Salt concentrations of 90 mM and 120 mM caused a significant decrease in K<sup>+</sup> and a very significant increase in Na<sup>+</sup> and Na/K in the sensitive genotypes compared with the control (Fig 5).

# 3.2. Chlorophyll and carotenoids contents

Salinity, genotypes, and their interaction were tested using ANOVA. Fisher's least significant difference test ( $P \le 0.05$ ) was used to evaluate the comparison of the relevant interactions. The statistically significant differences by different letters for sensitive genotypes and tolerant genotypes shown in Tables 3 and 4, respectively. Different letters among genotypes in the three salinity levels (One-way ANOVA) indicate significant differences. Salt treatment reduced the chlorophyll content of salt-sensitive genotypes (Table 3) more than that of salt-tolerant genotypes (Table 4).

Since carotenoids and chlorophylls are key components of the photosynthetic machinery, their role in light energy collection, membrane stability, and energy transfer has been intensively studied. In general, the levels of chlorophyll (*a* and *b*), total chlorophyll, and carotenoids were lower in non-stressed plants than in salt-stressed plants compared to the control. However, plants treated with 120 mM salinity were significantly different from control plants. This

| Genotypes | Salinity levels | Chlorophyll <i>a</i> (mg g <sup>-1</sup> FW) | Chlorophyllb (mg g <sup>-1</sup> FW) | Total Chlorophyll (mg g <sup>-1</sup> FW) | Carotenoids (mg g <sup>-1</sup> FW) |
|-----------|-----------------|--|--------------------------------------|---|-------------------------------------|
| 77        | Control         | 21.81±0.52a                                  | 9.30±0.66a                           | 27.17±0.65a                               | 98.47±6.96b                         |
|           | 90 mM           | 18.77±0.53b                                  | 6.68±0.14b                           | 23.50±0.66b                               | 103.05±2.19a                        |
|           | 120 mM          | 12.98±0.40c                                  | 4.97±0.49c                           | 16.24±0.51c                               | 57.84±5.73c                         |
| 81        | Control         | 17.01±1.20a                                  | 5.57±0.13a                           | 21.43±1.52a                               | 63.51±4.49a                         |
|           | 90 mM           | 15.58±0.33b                                  | 5.34±0.15a                           | 19.60±0.42b                               | 61.74±5.24b                         |
|           | 120 mM          | 11.78±1.17c                                  | 5.31±0.17a                           | 14.62±1.45c                               | 57.76±2.37c                         |
| 101       | Control         | 19.92±1.41b                                  | 6.28±0.44b                           | 25.14±0.71b                               | 86.22±2.07b                         |
|           | 90 mM           | 24.23±0.51a                                  | 8.35±0.18a                           | 30.47±1.29a                               | 91.19±2.58a                         |
|           | 120 mM          | 11.93±1.18c                                  | 6.31±0.62b                           | 14.68±0.21c                               | 72.60±2.26c                         |
| 120       | Control         | 19.15±1.08a                                  | 11.15±0.79a                          | 23.42±1.32a                               | 95.676±2.71a                        |
|           | 90 mM           | 14.27±1.31b                                  | 7.50±0.16b                           | 17.56±1.61b                               | 76.28±3.24b                         |
|           | 120 mM          | 12.61±0.98c                                  | 5.85±0.58c                           | 15.63±1.22c                               | 79.23±1.12b                         |
| 124       | Control         | 19.27±0.55a                                  | 8.88±0.50a                           | 23.90±1.69a                               | 101.74±7.19a                        |
|           | 90 mM           | 16.67±0.71b                                  | 6.08±0.56b                           | 20.92±0.44b                               | 89.29 ±1.89b                        |
|           | 120 mM          | 14.43±0.20c                                  | 4.80±0.37c                           | 18.18±1.80c                               | 29.37±2.91c                         |
| 126       | Control         | 15.60±0.88b                                  | 5.81±0.41a                           | 19.57±1.38b                               | 69.27±1.96b                         |
|           | 90 mM           | 19.77±1.82a                                  | 5.70±0.12a                           | 25.03±0.53a                               | 90.79±3.85a                         |
|           | 120 mM          | 14.43±1.12b                                  | 4.38±0.43a                           | 17.06±1.69c                               | 56.36±0.80c                         |
| 127       | Control         | 18.56±1.31a                                  | 5.56±0.39a                           | 23.48±1.66a                               | 66.72±4.72a                         |
|           | 90 mM           | 12.74±0.27b                                  | 4.20±0.36ab                          | 16.06±1.36c                               | 63.65±1.35 ab                       |
|           | 120 mM          | 17.54±1.74a                                  | 3.11±0.13b                           | 22.50±0.92b                               | 61.18±6.06b                         |
| 204       | Control         | 15.38±1.09b                                  | 5.60±0.40ab                          | 19.31±0.55b                               | 67.70±4.79b                         |
|           | 90 mM           | 18.76±0.40a                                  | 6.85±0.15a                           | 23.55±1.00a                               | 98.23±8.34a                         |
|           | 120 mM          | 13.42±1.33c                                  | 7.48±0.74a                           | 16.46±0.23c                               | 16.33±0.67c                         |
| 210       | Control         | 19.30±1.36a                                  | 5.91±0.17a                           | 24.39±1.72a                               | 79.84±2.26a                         |
|           | 90 mM           | 16.40±0.35b                                  | 6.21±0.26a                           | 20.55±1.74b                               | 80.30±3.41a                         |
|           | 120 mM          | 13.05±1.29c                                  | 4.91±0.07b                           | 16.35±0.67c                               | 19.53±0.28b                         |
| 213       | Control         | 16.49±1.17b                                  | 6.97±0.20a                           | 20.55±1.45b                               | 85.91±6.07b                         |
|           | 90 mM           | 18.56±0.39a                                  | 6.93±0.29a                           | 23.26±0.49a                               | 90.7±1.92a                          |
|           | 120 mM          | 11.09±1.10c                                  | 6.03±0.09a                           | 13.62±1.35c                               | 12.98±1.28c                         |

According to Fisher's least significant difference test at  $\alpha = 0.05$ , different letters indicate a significant difference within each genotype. The average and standard deviation are shown by vertical bars (two repetitions,  $P \le 0.05$ ). Where FW–Fresh weight.

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| Genotypes | Salinity levels | Chlorophyll <i>a</i> (mg g <sup>-1</sup> FW) | Chlorophyll <i>b</i> (mg g <sup>-1</sup> FW) | Total Chlorophyll (mg g <sup>-1</sup> FW) | Carotenoids (mg g <sup>-1</sup> FW) |
|-----------|-----------------|--|--|---|-------------------------------------|
| 2         | Control         | 16±1.13a                                     | 5.8±0.04a                                    | 21.58±1.53a                               | 67.40±4.77b                         |
|           | 90 mM           | 17±0.36a                                     | 5.6±0.08a                                    | 22.36±0.47a                               | 72.43±1.54a                         |
|           | 120 mM          | 14.8±1.47b                                   | 5.3±0.11a                                    | 20.86±2.07a                               | 53.30±5.28c                         |
| 11        | Control         | 20.3±1.15a                                   | 6.2±0.15a                                    | 26.25±1.48a                               | 89.05±5.04ab                        |
|           | 90 mM           | 21.8±2.00a                                   | 5.3±0.15a                                    | 27.51±2.53a                               | 93.79±8.62a                         |
|           | 120 mM          | 16.68±1.30b                                  | 5.6±0.17a                                    | 22.33±1.74b                               | 86.68±6.74ab                        |
| 86        | Control         | 17±1.20a                                     | 7.6±0.05a                                    | 24.96±1.76a                               | 78.24±5.53b                         |
|           | 90 mM           | 18.6±0.39a                                   | 6.3±0.09a                                    | 24.86±0.53a                               | 82.84±1.76a                         |
|           | 120 mM          | 15.2±1.50b                                   | 6.15±0.13a                                   | 21.535±2.13c                              | 66.67±6.60c                         |
| 109       | Control         | 18.37±1.04b                                  | 7.2±0.24a                                    | 25.657±1.45a                              | 68.58±3.88ab                        |
|           | 90 mM           | 21.2±1.95a                                   | 6.15±0.35a                                   | 27.435±2.52a                              | 73.50±6.76a                         |
|           | 120 mM          | 18.5±1.44b                                   | 5.8±0.41a                                    | 24.23±1.88b                               | 65.05±5.06b                         |
| 151       | Control         | 17.3±1.22a                                   | 7±0.05a                                      | 24.53±1.73a                               | 78.58±5.56a                         |
|           | 90 mM           | 17.9±0.38a                                   | 6.65±0.09a                                   | 24.55±0.52a                               | 79.75±1.69a                         |
|           | 120 mM          | 16.89±1.67a                                  | 5.83±0.12a                                   | 22.72±2.25a                               | 75.01±7.43a                         |
| 191       | Control         | 17.2±0.97a                                   | 8.21±0.62a                                   | 25.541±1.44a                              | 82.16±4.65b                         |
|           | 90 mM           | 19±1.75a                                     | 7.2±0.61a                                    | 26.2±2.41a                                | 96.573±8.88a                        |
|           | 120 mM          | 16.8±1.31a                                   | 6.5±0.60a                                    | 23.63±1.77a                               | 77.35±6.02b                         |
| 199       | Control         | 17.78±1.26a                                  | 7.19±0.71a                                   | 24.97±0.53a                               | 76.54±5.41a                         |
|           | 90 mM           | 17.9±0.38a                                   | 6.86±0.29a                                   | 24.76±2.30a                               | 81.47±1.73a                         |
|           | 120 mM          | 16.98±1.68a                                  | 6.21±0.22a                                   | 23.19±1.40a                               | 73.14±7.24ab                        |
| 205       | Control         | 17.38±0.98a                                  | 7.31±0.05a                                   | 24.69±2.23a                               | 71.493±4.04a                        |
|           | 90 mM           | 18.01±1.66a                                  | 6.3±0.09a                                    | 24.31±1.77a                               | 73.480±6.75a                        |
|           | 120 mM          | 16.8±1.31a                                   | 5.9±0.13a                                    | 22.7±0.66a                                | 71.47±5.56a                         |
| 232       | Control         | 13.8±0.39a                                   | 9±0.45a                                      | 23.18±1.01a                               | 79.448±2.25ab                       |
|           | 90 mM           | 14.2±0.60a                                   | 8.8±0.50a                                    | 23.8±0.28a                                | 86.16±3.66a                         |
|           | 120 mM          | 12.6±0.18a                                   | 6.71±0.43ab                                  | 19.63±1.96b                               | 76.1±1.08b                          |
| 239       | Control         | 18.68±1.32a                                  | 8.6±0.61a                                    | 27.73±0.59a                               | 79.38±5.61ab                        |
|           | 90 mM           | 19.1±0.41a                                   | 8.23±0.70a                                   | 27.63±2.53a                               | 82.97±1.76a                         |
|           | 120 mM          | 17.6±1.74a                                   | 7.51±0.31a                                   | 25.51±1.84a                               | 65.66±6.50b                         |

#### Table 4. Effect of various concentrations of NaCl after 21 days on chlorophyll contents in leaves of 10 tolerant genotypes of wheat.

According to Fisher's least significant difference test at  $\alpha = 0.05$ , different letters indicate a significant difference within each genotype. The average and standard deviation are shown by vertical bars (two repetitions,  $P \le 0.05$ ). Where FW–Fresh weight.

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difference was not significant in the tolerant genotypes. In contrast, the tolerant genotypes had higher chlorophyll and carotenoid contents than the sensitive genotypes at both salinity levels at all stages. This was probably due to inhibition of ribulose-1,5-bisphosphate enzyme and the structural destruction of the chloroplast and photosynthetic apparatus, ultimately resulting in a decrease in photosynthetic pigments such as chlorophyll, carotenoids, and CSI [61,62]. Similar results were also observed in wheat plants under high salt stress, with reductions in photosynthetic pigments, photosynthetic rate, stomatal conductance, and  $CO_2$  intake [63].

This study showed a very significant reduction in carotenoid content when treated with 90 mM and 120 mM salt stress in the sensitive genotypes. In contrast, these changes were insignificant in the tolerant genotypes. Carotenoids are responsible for quenching singlet oxygen [64]. Therefore, their comparable content in a cultivar may determine its relative tolerance. *Hamada* [65] reported that the decrease in chlorophyll content was due to the increase in chlorophyll enzyme activity and the instability of protein complexity of pigments. Carotenoids are

important for plants under stress because they play a significant role as precursors in signaling during plant expansion under environmental stress, so they are very important for photoprotection of photosynthesis [66]. This result is confirmed by *Shah*, *et al.* [67]. They also found that chlorophyll and carotenoids decrease with reduced salinity. The decrease in pigment amounts and ratios in this study is also consistent with the results of *Pastuszak*, *et al.* [68], who observed that salt-tolerant wheat genotypes produced less chlorophyll *a*, chlorophyll *b*, carotenoids, and lower chlorophyll *a* to *b* ratio with increasing NaCl.

# 3.3. Na<sup>+</sup> and K<sup>+</sup>

One of the biggest global issues negatively impacting agricultural yields is salinity. Water stress, cytotoxicity caused by excessive uptake of ions such as sodium (Na<sup>+</sup>) and chloride (Cl<sup>-</sup>), and imbalanced nutrient ratios hinder plant growth and development. Due to the production of reactive oxygen species (ROS), this is also often accompanied by oxidative stress [69].

In this study, salt caused a decrease in  $K^+$  content in all wheat genotypes (Figs 4 and 5). The  $K^+$  deficiency was caused by the presence of too much Na<sup>+</sup> in the growth media, which is known to have a negative effect on  $K^+$  absorption in plants [70]. Due to the role of  $K^+$  in osmotic control and competition with Na<sup>+</sup> [71], salt tolerance is correlated with  $K^+$  concentration [72]. In general, plants control  $K^+$  uptake and stop Na<sup>+</sup> efflux into the cell by maintaining the optimal K/Na ratio in the cytosol. The transport mechanisms involved in the use of Na<sup>+</sup> as an osmotic solution have received the most attention in research evaluating plant responses to salt stress [73]. Sodium and potassium compete for absorption through a common transport pathway. This competition is successful because Na<sup>+</sup> concentration is often much higher than K<sup>+</sup> concentration in saline conditions. Moreover, it has been argued that most plants are susceptible to salt stress because we are unable to keep Cl and Na<sup>+</sup> out of transpiration currents [74].

Therefore, plants could show greater tolerance by limiting the uptake of harmful ions or maintaining normal nutrient ion levels, which has been done. Uptake mechanisms that discriminate similar ions such as Na<sup>+</sup> and K<sup>+</sup> could be critical selection factors for salt tolerance in wheat genotyping and breeding for best nutrient uptake under salt stress.

# 3.4 Indicator of cellular oxidation degree of thiobarbituric acid reactive material (TBARM)

Lipids are the main source of energy for cells [75]. They are the basic mechanism for cell membrane assembly and act as a sensitive insulator between organs and hormones [76]. Salinity resistance and water stress are related to the two lipid layers containing unsaturated fatty acids, which maintain membrane stability in the presence of sugar stress (trehalose. Membrane lipid peroxidation is a combination of malondialdehyde (MDA), propane, butanal, hexane, heptanal, and propanedimethylacetal. These examples are used to determine the amount of membrane lipid peroxidation; an increase in lipid peroxidation is considered to indicate greater oxidative stress. Lipid peroxidation of membranes is a sign of oxidative stress. At the same time, the TBARM assay, which measures malondialdehyde, can be evaluated as an indicator of oxidation levels at the cellular and molecular levels [77].

The application of 120 mM showed a significant downward trend in susceptible genotypes (Table 5), but in resistant genotypes, this index continued to increase from 90 mM to 120 mM with increasing salt stress (Table 6). Since this oxidation index is measurable at the cellular and molecular levels, it can be said that the decreasing trend of this index indicates cell death.

| Genotypes | Salinity levels | TBARM (mg MAD kg <sup>-1</sup> ) | LOX (µmol min <sup>-1</sup> g <sup>-1</sup> ) | Protein (mg g <sup>-1</sup> FW) | Proline (µmol L <sup>-1</sup> ) |
|-----------|-----------------|----------------------------------|---|---------------------------------|---------------------------------|
| 77        | Control         | 2.5±0.07c                        | 3.8±0.07c                                     | 1.5±0.04c                       | 1.3±0.04c                       |
|           | 90 mM           | 7.8±0.06b                        | 4.3±0.14b                                     | 4.4±0.07a                       | 4.1±0.14a                       |
|           | 120 mM          | 9.2±0.07a                        | 5±0.21a                                       | 3.5±0.07b                       | 3.7±0.20b                       |
| 81        | Control         | 3.3±0.04c                        | 2.9±0.07c                                     | 2.5±0.06c                       | 1.45±0.15c                      |
|           | 90 mM           | 6.9±0.07b                        | 4.7±0.14a                                     | 3.6±0.08b                       | 2.4±022b                        |
|           | 120 mM          | 10±0.14a                         | 4±0.21b                                       | 4.3±0.42a                       | 3.8±0.15a                       |
| 101       | Control         | 3.5±0.07c                        | 3.3±0.14c                                     | 3.2±0.14c                       | 0.9±0.08c                       |
|           | 90 mM           | 7.2±0.14b                        | 4±0.14b                                       | 4.4±0.14a                       | 2.2±0.07b                       |
|           | 120 mM          | 12.5±0.14a                       | 4.5±0.21a                                     | 3.9±0.15b                       | 3.1±0.13a                       |
| 120       | Control         | 3±0.07c                          | 4±0.21b                                       | 2.8±0.04b                       | 1.2±0.14c                       |
|           | 90 mM           | 9±0.14b                          | 4.2±0.28ab                                    | 3.8±0.21a                       | 3.8±0.15a                       |
|           | 120 mM          | 12±0.15a                         | 4.4±0.14a                                     | 4.2±0.28a                       | 3.2±0.16b                       |
| 124       | Control         | 2.9±0.06c                        | 3.4±0.07b                                     | 1.5±0.05c                       | 1.54±0.17c                      |
|           | 90 mM           | 8.4±0.16b                        | 5±0.28a                                       | 3.4±0.06b                       | 3.9±0.18a                       |
|           | 120 mM          | 11±0.15a                         | 4.5±0.23a                                     | 3.6±0.08a                       | 3.5±0.15b                       |
| 126       | Control         | 3.7±0.07c                        | 2.8±0.21c                                     | 1.1±0.04c                       | 1.32±0.07c                      |
|           | 90 mM           | 7.9±0.13b                        | 5.9±0.20a                                     | 2.9±0.09a                       | 3±0.12b                         |
|           | 120 mM          | 11.1±0.12a                       | 5.1±0.28b                                     | 2.4±0.04b                       | 3.9±0.05a                       |
| 127       | Control         | 2.56±0.06c                       | 3.8±0.13b                                     | 2.4±0.07c                       | 1.7±0.05c                       |
|           | 90 mM           | 7.4±0.08b                        | 6.2±0.15a                                     | 4.3±0.04a                       | 4±0.07a                         |
|           | 120 mM          | 10.5±0.09a                       | 5.7±0.17a                                     | 3.5±0.07b                       | 3.7±0.15b                       |
| 204       | Control         | 3.17±0.03c                       | 3.1±0.13c                                     | 3.1±0.07c                       | 1.1±0.09c                       |
|           | 90 mM           | 8.4±0.07b                        | 7±0.27a                                       | 4.5±0.07a                       | 3.1±0.16b                       |
|           | 120 mM          | 10.8±0.14a                       | 5.4±0.21b                                     | 3.7±0.14b                       | 4±0.19a                         |
| 210       | Control         | 3.8±0.07c                        | 3.5±0.22c                                     | 3.3±0.14c                       | 1.25±0.07c                      |
|           | 90 mM           | 8±0.07b                          | 5.1±0.23a                                     | 5.1±0.33a                       | 2.44±0.12b                      |
|           | 120 mM          | 12.1±0.01a                       | 4.8±0.18b                                     | 4±0.25b                         | 3.3±0.17a                       |
| 213       | Control         | 3.21±0.07c                       | 4±0.12c                                       | 3.23±0.24b                      | 1.38±0.18c                      |
|           | 90 mM           | 7.9±0.14b                        | 6.1±0.20a                                     | 4.82±0.27a                      | 3±0.19b                         |
|           | 120 mM          | 10±0.16a                         | 5.3±0.21b                                     | 3.53±0.21b                      | 3.45±0.16a                      |

Table 5. Effect of various concentrations of NaCl after 21 days on TBARM, LOX, protein, and proline in leaves of 10 sensitive genotypes of wheat.

According to Fisher's least significant difference test at  $\alpha = 0.05$ , different letters indicate a significant difference within each genotype. The average and standard deviation are shown by vertical bars (two repetitions,  $P \le 0.05$ ). Where MAD–malonaldehyde, FW–Fresh weight.

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# 3.5. Lipoxygenase (LOX)

One of the most important enzyme systems at the interface by changing the lipid of cell membranes is the LOX enzyme system [78]. This enzyme controls the reaction of the connection between oxygen molecules and unsaturated fatty acids and the formation of unsaturated fatty acid hydroxides [79]. The oxidation of fatty acids by the activity of this enzyme leads to the formation of oxygen free radicals [80].

According to the results of this study, the effect of salt stress on the accumulation of LOX was significant (Tables 5 and 6), indicating that salt stress has a great effect on the oxidative property of LOX. A high LOX index indicates the abundance of reactive oxygen radicals and the intensity of the oxidative process. In general, stress increases the level of oxygen free radicals in plant cells, such as hydrogen peroxide, and the resulting hydrogen peroxide increases the level of the enzyme LOX in plant cells [81]. This enzyme catalyzes unsaturated and long-chain fatty acids containing a cis bond. Linoleic acid and linolenic acid are the most highly

| Genotypes | Salinity levels | TBARM (mg MAD kg <sup>-1</sup> ) | LOX (µmol min <sup>-1</sup> g <sup>-1</sup> ) | Protein (mg g <sup>-1</sup> FW) | Proline (µmol L <sup>-1</sup> ) |
|-----------|-----------------|----------------------------------|---|---------------------------------|---------------------------------|
| 2         | Control         | 2.2±0.14c                        | 4.2±0.28c                                     | 1.1±0.04c                       | 2±0.05c                         |
|           | 90 mM           | 5.4±0.07b                        | 8±0.35b                                       | 4.9±0.24b                       | 4.1±0.15b                       |
|           | 120 mM          | 7.2±0.14a                        | 9.8±0.14a                                     | 6.1±0.30a                       | 5.8±0.25a                       |
| 11        | Control         | 3.1±0.14b                        | 3.7±0.14c                                     | 2.8±0.21c                       | 2.5±0.14c                       |
|           | 90 mM           | 4.9±0.14a                        | 6.9±0.07b                                     | 5.1±0.27b                       | 4.5±0.16b                       |
|           | 120 mM          | 5.2±0.28a                        | 10.7±0.99a                                    | 6.9±0.28a                       | 5.2±0.23a                       |
| 86        | Control         | 3.2±0.07c                        | 3.8±0.07c                                     | 3.2±0.19c                       | 2.5±0.12c                       |
|           | 90 mM           | 4.8±0.07b                        | 7.8±0.07b                                     | 4.8±0.21b                       | 5.5±0.27b                       |
|           | 120 mM          | 5.5±0.14a                        | 11.4±0.57a                                    | 7.1±0.20a                       | 6.4±0.30a                       |
| 109       | Control         | 2.7±0.04c                        | 4.7±0.42c                                     | 3.7±0.18c                       | 2.75±0.09c                      |
|           | 90 mM           | 4.9±0.03b                        | 7.2±0.28b                                     | 5.5±0.26b                       | 5.1±0.17b                       |
|           | 120 mM          | 6.6±0.06a                        | 12.2±0.28a                                    | 6.1±0.30a                       | 7.8±0.33a                       |
| 151       | Control         | 2.4±0.08c                        | 5±0.07c                                       | 2.9±0.14c                       | 2.2±0.08c                       |
|           | 90 mM           | 5.3±0.07b                        | 7.4±0.07b                                     | 6±0.25b                         | 7±0.36a                         |
|           | 120 mM          | 5.8±0.04a                        | 9.9±0.14a                                     | 7.65±0.32a                      | 6.5±0.32b                       |
| 191       | Control         | 3.1±0.14c                        | 4.5±0.21c                                     | 4±0.20b                         | 3±0.14c                         |
|           | 90 mM           | 4.7±0.07b                        | 6.8±0.14b                                     | 7.1±0.35a                       | 6.5±0.30a                       |
|           | 120 mM          | 6.4±0.07a                        | 11.7±0.42a                                    | 6.8±0.36a                       | 5.2±0.28b                       |
| 199       | Control         | 3±0.07c                          | 3.9±0.21c                                     | 1.9±0.13c                       | 1.9±0.15c                       |
|           | 90 mM           | 5±0.14b                          | 8.5±0.21b                                     | 5.4±0.16b                       | 6.5±0.28a                       |
|           | 120 mM          | 8.4±0.07a                        | 12.1±0.14a                                    | 6.4±0.23a                       | 5.8±0.24b                       |
| 205       | Control         | 2.8±0.04c                        | 4.1±0.14c                                     | 2.7±0.10b                       | 1.4±0.08c                       |
|           | 90 mM           | 5.1±0.14b                        | 8.2±0.28b                                     | 6.7±0.27a                       | 4.7±0.21b                       |
|           | 120 mM          | 5.4±0.07a                        | 10.4±0.07a                                    | 7±0.34a                         | 5.9±0.22a                       |
| 232       | Control         | 3.1±0.07c                        | 4±0.14c                                       | 2.4±0.014c                      | 2.1±0.14c                       |
|           | 90 mM           | 4.9±0.06b                        | 6.9±0.07b                                     | 5.8±0.25b                       | 7.2±0.36a                       |
|           | 120 mM          | 5.7±0.07a                        | 12.6±0.28a                                    | 6.7±0.28a                       | 6.8±0.29b                       |
| 239       | Control         | 3.5±0.10c                        | 4.5±0.14c                                     | 3.5±0.18c                       | 1.25±0.10c                      |
|           | 90 mM           | 5.3±0.04b                        | 7.2±0.07b                                     | 6.1±0.29b                       | 5.4±0.23b                       |
|           | 120 mM          | 6.9±0.04a                        | 10.8±0.07a                                    | 7.2±0.33a                       | 7.1±035a                        |

Table 6. Effect of various concentrations of NaCl after 21 days on TBARM, LOX, protein, and proline in leaves of 10 tolerant genotypes of wheat.

According to Fisher's least significant difference test at  $\alpha = 0.05$ , different letters indicate a significant difference within each genotype. The average and standard deviation are shown by vertical bars (two repetitions,  $P \le 0.05$ ). Where MAD–malonaldehyde, FW–Fresh weight.

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unsaturated fatty acids in plant cell structure, providing an ideal starting material for the activity of this enzyme [82]. Toxic concentrations of reactive oxygen radicals cause severe damage to protein structures, inhibition of the activity of various enzymes in metabolic pathways, and consequent oxidation of macromolecules such as lipids and DNA. The exacerbation and persistence of these adverse events can lead to cell death [83].

# 3.6. Protein

The results of data analysis showed that salinity stress in the Tables 5 and 6 and its interaction with salinity stress were very significant for protein adjectives. Protein synthesis changes in response to environmental stresses such as salt, heat shock, anaerobic conditions, drought, osmotic shock, wounding, and cold stress [84]. Such stresses increased the synthesis of some proteins and decreased the synthesis of others. It appears that the proteins induced by salinity effectively tolerated this stress.

According to the results of average protein comparisons, protein levels increased with increasing salinity stress in all genotypes (Tables 5 and 6). However, with increasing salinity up to 120 mM, this increase was not stable in the susceptible genotypes, so that in these genotypes the level of spreading decreased from 90 mM to 120 mM with increasing salt stress (Table 5). However, an increasing trend was still observed in the resistant genotypes (Table 6). Proteins that accumulate in plants under salinity stress serve as nitrogen reserves in osmotic regulation. In response to salinity stress, proteins can be newly formed or institutionalized in concentration [84]. There are low concentrations, and when plants are exposed to repair and repair damage and salt stress, their concentration increases [85]. In osmotic or ionic stress, elevated stress proteins are essential for cell survival at the level of metabolic inhibition [86]. Other osmotic and physiological adaptations, such as changes in root and shoot development and transpiration, may be involved in these tactics, suggesting that proteins are resynthesized in plants under salt stress [87].

Salt stress leads to quantitative and qualitative changes in the amount of soluble proteins [88]. Proteins that increase in plants under salt stress may be a form of nitrogen storage that is later used by the plant [89]. Apples may play a role in osmotic adaptation, such as the production of pseudo-osmotic proteins or proteins, or they may alter the structure of the cell wall. These proteins may be synthesized in response to salt stress, or perhaps there are structures of concentrations low and high [90].

In this study, the concentration of leaf soluble proteins increased with increasing salt concentration in all genotypes (Tables 5 and 6). It can be said that antiperspirants increase antioxidant activity and also prevent protein degradation, thus increasing protein levels. Many proteins induced by salt stress are molecular chaperones, and some of them are synthesized to counteract the oxidative stress caused by salinity and prevent the destruction of structural and functional proteins by oxidative stress [91].

However, this changed in the sensitive genotypes, so that the control group had the lowest protein level and the salinity group had the highest protein level, 90 mM (Table 5). The significant reduction in protein content in the sensitive genotypes can be attributed to both protein degradation and reduced protein synthesis. Decreased water potential in leaves appears to lead to a sharp decrease in polyribosomes and monoribosomes, implying a decrease in protein synthesis. Oxygen-free radicals, which have a high composition of proteins, also lead to their oxidation [92].

#### 3.7. Proline

The mean comparison test (Tables 5 and 6) shows that with increasing salinity, proline accumulation increases in all genotypes.

In plant stress physiology, the accumulation of compatible solutes is generally believed to play a role in maintaining osmotic balance in cells [93,94]. For example, proline accumulation in plants increases salt resistance [95]. As salinity increased from 90 mM to 120 mM, proline significantly decreased in susceptible genotypes (Table 5), whereas it continued to increase in resistant genotypes as salinity increased (Table 6). Therefore, the embryo may play the role of proline in protecting against osmosis, which increases during plant growth. It helps to stabilise the membrane and reduce the effects of NaCl on cell membrane degradation. It is important for regulating osmotic potential, removing free radicals, and preventing denaturation of macromolecules in cellular pH when proline is stressed. Proline also acts as a nitrogen and carbon source for plants under extreme stress and increases the stress tolerance of plants [96]. Therefore, it can be said that proline is the most effective regulator of osmotic pressure of higher plants under stress salinity. Under stress conditions, proline plays a role in maintaining

membrane structure, creating osmotic compatibility, and maintaining the structure of enzymes in the cell [97].

Therefore, genotypes that produce more proline may be more resistant to stress [98]. Considering that salt stress is one of the most important stress factors affecting plant performance, understanding the mechanisms that plants use when exposed to stress is of great importance for plant breeding research. Plants need to enhance resistance mechanisms such as reactive oxygen radical removal and cellular defence system to maintain equilibrium under stress [99].

# 4. Conclusions

Our study found that salinity had a significant effect on the photochemical reaction of the photosynthetic apparatus in wheat plants, as indicated by changes in parameters of chlorophyll-a fluorescence. Additionally, the results showed that salt stress had similar mechanisms of action on the light-dependent photosynthetic phase in nearly all genotypes, but certain wheat genotypes were particularly sensitive to salt stress when it comes to PSII. Salinity stress led to an increase in energy dissipation and damage to the oxygen evolving complex and reaction centers in plants. However, to cope with and survive this stress, plants boost the activity of antioxidant enzymes that break down or remove harmful compounds from the cell. As a result, their levels in the plant increase during salt stress.

Therefore, molecular studies need to be conducted and linked to morphological, physiological, and biochemical traits to understand the processes behind the effects of salt stress on plant growth and photosynthetic efficiency. Nevertheless, we recommend the use of prompt chlorophyll fluorescence parameters as bio-indicators for a quick survey of tolerance in wheat plants under salinity.

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