

RESEARCH ARTICLE

Optimization of the Ethanol Recycling Reflux Extraction Process for Saponins Using a Design Space Approach

Xingchu Gong, Ying Zhang, Jianyang Pan, Haibin Qu*

Pharmaceutical Informatics Institute, College of Pharmaceutical Sciences, Zhejiang University, Hangzhou, 310058, China

*quhb@zju.edu.cn



 OPEN ACCESS

Citation: Gong X, Zhang Y, Pan J, Qu H (2014) Optimization of the Ethanol Recycling Reflux Extraction Process for Saponins Using a Design Space Approach. PLoS ONE 9(12): e114300. doi:10.1371/journal.pone.0114300

Editor: Qinghui Zhang, University of Nebraska Medical Center, United States of America

Received: September 2, 2014

Accepted: November 8, 2014

Published: December 3, 2014

Copyright: © 2014 Gong et al. This is an open-access article distributed under the terms of the [Creative Commons Attribution License](http://creativecommons.org/licenses/by/4.0/), which permits unrestricted use, distribution, and reproduction in any medium, provided the original author and source are credited.

Data Availability: The authors confirm that all data underlying the findings are fully available without restriction. All relevant data are within the paper.

Funding: This work was supported by the National S&T Major Project of China (2012ZX09101201-003) and Research Fund for the Doctoral Program of Higher Education of China (20110101120149). Haibin Qu received the first fund. The website is (<http://www.most.gov.cn/>). Xingchu Gong received the second fund. The website is (http://www.cutech.edu.cn/cn/kjyj/gdxxbsdkyj/A010301index_1.htm). Funds were used to buy materials, chemicals, and pay labor costs. The funders had no role in study design, data collection and analysis, decision to publish, or preparation of the manuscript.

Competing Interests: The authors have declared that no competing interests exist.

Abstract

A solvent recycling reflux extraction process for *Panax notoginseng* was optimized using a design space approach to improve the batch-to-batch consistency of the extract. Saponin yields, total saponin purity, and pigment yield were defined as the process critical quality attributes (CQAs). Ethanol content, extraction time, and the ratio of the recycling ethanol flow rate and initial solvent volume in the extraction tank (RES) were identified as the critical process parameters (CPPs) via quantitative risk assessment. Box-Behnken design experiments were performed. Quadratic models between CPPs and process CQAs were developed, with determination coefficients higher than 0.88. As the ethanol concentration decreases, saponin yields first increase and then decrease. A longer extraction time leads to higher yields of the ginsenosides Rb₁ and Rd. The total saponin purity increases as the ethanol concentration increases. The pigment yield increases as the ethanol concentration decreases or extraction time increases. The design space was calculated using a Monte-Carlo simulation method with an acceptable probability of 0.90. Normal operation ranges to attain process CQA criteria with a probability of more than 0.914 are recommended as follows: ethanol content of 79–82%, extraction time of 6.1–7.1 h, and RES of 0.039–0.040 min⁻¹. Most of the results of the verification experiments agreed well with the predictions. The verification experiment results showed that the selection of proper operating ethanol content, extraction time, and RES within the design space can ensure that the CQA criteria are met.

Introduction

Solvent recycling reflux extraction is an extraction process wherein extraction and concentration are conducted simultaneously [1]. The extract in the extraction tank is pumped out and concentrated in the concentration tank during the extraction process; meanwhile, the evaporated solvent is condensed and pumped back into the extraction tank. Compared with the conventional heat reflux extraction, solvent recycling reflux extraction has several advantages [2]. Because the solvent is renewed in the extraction, the mass transfer driving force is greater, which leads to a shorter extraction time. The reuse of the solvent in the extraction also decreases the amount of solvent needed. In addition, no storage tank is required before the concentration process, which can reduce the fixed investment. The high efficiency of the solvent recycling reflux extraction process was verified in the extraction of polysaccharides from *Grifola frondosa* [3] and the preparation of Jianwei Xiaozhang pills [4], Xinmaikang tablets [1], and other botanical extracts [2]. Because of these advantages, solvent recycling reflux extraction is increasingly applied to extract botanical components in botanical medicine factories to lower costs.

In addition to economic considerations, botanical drug quality is important for botanical drug manufacturers. Because of the complexity of the compositions of botanical extracts, maintaining batch-to-batch consistency is a challenging task. Quality by design (QbD) is a paradigm that has recently been used to improve the batch-to-batch consistency of the pharmaceutical process based on risk management and knowledge management [5,6]. In the implementation of the QbD concept, steps including critical quality attribute (CQA) definition, risk assessment, critical process parameter (CPP) determination, design space development, control strategy design, and continual improvement in the product lifecycle are required [7,8]. The design space is a region for the control of process parameters. When the parameters vary within this region, the change in the process product quality can be neglected [7]. To determine the design space, mathematical models between CQAs and CPPs are required. Experimental design is often applied to establish the models [9]. To quantify the ability of the design space to keep the CQAs within the desired ranges, the probability of attaining the CQA ranges must be calculated [10]. Monte-Carlo and Bayesian methods are commonly used to calculate this probability [11–13]. Recently, ethanol precipitation and water precipitation, two separation processes that are widely applied in the manufacturing of botanical drugs [14,15], have been successfully optimized according to the QbD paradigm.

To realize a solvent recycling reflux extraction process with high batch-to-batch consistency, the extraction of *Panax notoginseng* was investigated. *Panax notoginseng*, the root of *Panax notoginseng* (Burk.) F. H. Chen, is a medicinal and edible plant in China and is used as a dietary supplement in the USA [16]. Many botanical drugs widely applied in China are made from *Panax notoginseng*, such as Xuesaitong injections, Compound Danshen Dripping Pills, and Yunnan Baiyao. In this work, the extraction process was optimized using a design space approach

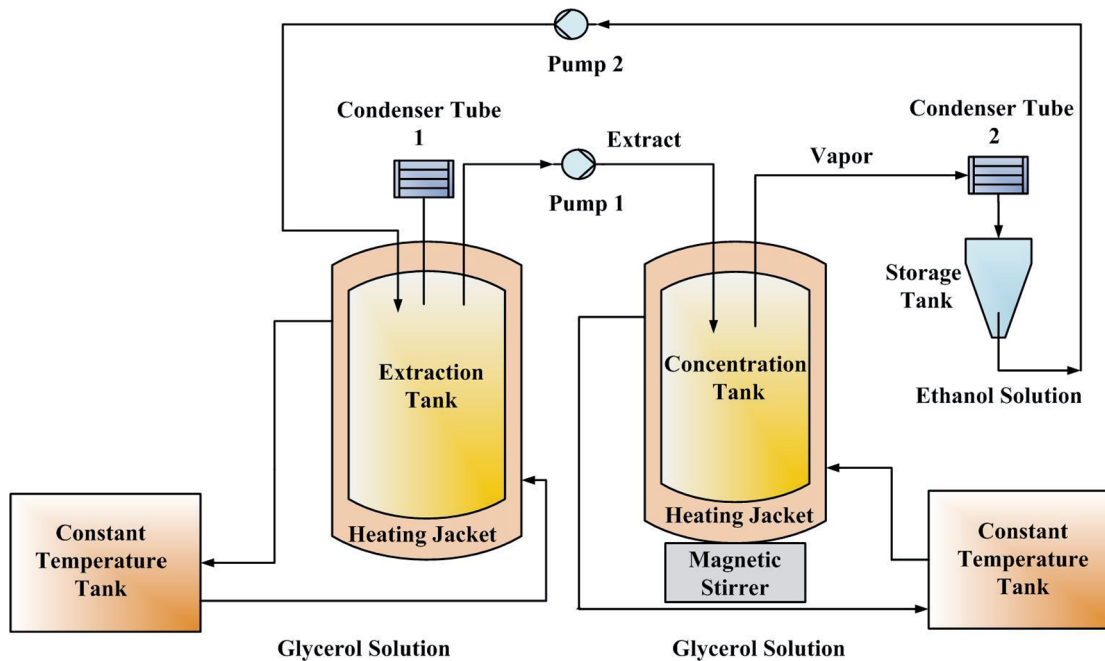


Figure 1. Schematic chart of the experimental setup.

doi:10.1371/journal.pone.0114300.g001

[17] consisting of CQA definition, risk assessment, CPP determination, design space development and verification. The CQAs of the extraction process were defined, and the CPPs were identified via risk assessment. Quantitative models were developed between CQAs and CPPs. The influences of different parameters were discussed. The probability-based design space was calculated using a Monte-Carlo method. Finally, the design space was verified.

Methods and Materials

Materials and chemicals

Panax notoginseng was collected from Wenshan of Yunnan Province (China). No specific permissions were required for the described field studies. The locations are neither privately owned nor protected by the Chinese government. No endangered or protected species were sampled. The specific location of this study is longitude: 120.07E, latitude: 30.28N. Standards of the notoginsenoside R₁, ginsenoside Rg₁, ginsenoside Rb₁, and ginsenoside Rd were purchased from Shanghai Winherb Pharmaceutical Technology Development Co., Ltd. (Shanghai, China). Acetonitrile (HPLC grade) and methanol (HPLC grade) were obtained from Merck (Darmstadt, Germany). Formic acid (HPLC grade) was purchased from Tedia (Darmstadt, Germany). Ethanol (analytical grade) was purchased from Shanghai Lingfeng Chemical Reagent Co., Ltd. (Shanghai, China). Tartrazine (HPLC grade) was purchased from Aladdin Industrial Corporation

Table 1. Coded and uncoded values of factors.

Parameters	Symbols	Coded values		
		-1	0	1
Ethanol concentration (v/v, %)	X ₁	70	80	90
Extraction time (h)	X ₂	4	7	10
RES (min ⁻¹)	X ₃	0	0.02	0.04

doi:10.1371/journal.pone.0114300.t001

(Shanghai, China). Glycerol (analytical grade) was purchased from China Sun Specialty Products Co., Ltd. (Changshu, China). A Milli-Q academic water purification system (Milford, MA, USA) was used to produce deionized water.

Procedures

A schematic chart of the experimental setup is shown in [Figure 1](#). Two constant-temperature tanks (ZCY-15B, Ningbo Tianheng Instrument Factory, Ningbo, China) were used to heat the extraction tank and concentration tank. The extract in the extraction tank was pumped into the concentration tank. The solvent was evaporated in the concentration tank and then condensed. The ethanol solution was collected in a storage tank and then pumped back into the extraction tank with a fixed flow rate. In this process, the saponins were kept in the concentration tank, and the solvent was recycled. In the experiments, 50.0 g of *Panax notoginseng* and 500.0 ml of ethanol-water mixture were added to the extraction tank. After soaking for 2 h, 150 mL of extract was pumped into the concentration

Table 2. Box-Behnken design experiments and results.

Run order	X ₁	X ₂	X ₃	Saponin yield(mg/g <i>Panax notoginseng</i>)				TSP (%)	Pigment yield (mg/g <i>Panax notoginseng</i>)
				R ₁	Rg ₁	Rb ₁	Rd		
1	70	7	0.04	13.56	43.96	45.06	10.16	25.34	2.64
2	70	7	0	11.39	42.01	41.10	8.65	29.89	0.39
3	90	7	0	10.14	33.17	30.42	6.84	29.78	0.14
4	70	10	0.02	11.19	38.24	42.69	9.24	23.11	7.87
5	80	4	0	11.04	38.18	35.93	7.99	28.96	0.23
6	80	4	0.04	12.39	44.03	43.04	9.53	27.29	1.81
7	80	7	0.02	13.26	46.34	44.41	9.54	27.34	2.92
8	80	7	0.02	13.27	45.60	45.39	9.40	27.23	2.88
9	80	10	0.04	14.91	48.14	50.71	10.45	27.25	3.68
10	90	4	0.02	10.81	35.58	33.42	7.65	25.68	0.75
11	90	10	0.02	11.15	37.46	35.17	7.72	25.89	3.02
12	80	10	0	11.47	43.46	43.09	9.53	32.55	0.35
13	80	7	0.02	14.40	47.00	47.75	9.67	27.61	3.41
14	90	7	0.04	10.82	37.79	36.27	8.02	32.03	1.13
15	70	4	0.02	11.80	42.99	43.52	8.97	25.11	4.86
16	80	7	0.02	13.77	46.11	47.36	9.28	27.32	3.11

doi:10.1371/journal.pone.0114300.t002

Table 3. Conditions and results of verification experiments.

Verification experiment No.		V ₁	V ₂
Ethanol concentration (%)		80	90
Extraction time (h)		7	4
RES (min ⁻¹)		0.04	0.04
Probability		0.97	0
Within design space		Yes	No
Notoginsenoside R ₁ yield (mg/g <i>Panax notoginseng</i>)	EV	14.04 ± 0.20	11.01 ± 0.32
	PV	14.14	9.60
	RD (%)	0.73	12.76
Ginsenoside Rg ₁ yield (mg/g <i>Panax notoginseng</i>)	EV	46.45 ± 0.77	36.54 ± 1.16
	PV	47.33	35.22
	RD (%)	1.89	3.61
Ginsenoside Rb ₁ yield(mg/g <i>Panax notoginseng</i>)	EV	47.23 ± 0.17	33.77 ± 0.82
	PV	47.53	33.10
	RD (%)	0.64	1.98
Ginsenoside Rd yield(mg/g <i>Panax notoginseng</i>)	EV	9.94 ± 0.17	7.53 ± 0.14
	PV	10.08	7.92
	RD (%)	1.40	5.22
TSP (%)	EV	28.68 ± 0.26	30.57 ± 0.53
	PV	29.19	29.84
	RD (%)	1.79	2.40
Pigment yield(mg/g <i>Panax notoginseng</i>)	EV	1.85 ± 0.20	0.57 ± 0.05
	PV	2.12	0.62
	RD (%)	14.57	8.40

EV: Experimental value

PV: Predicted value.

doi:10.1371/journal.pone.0114300.t003

tank at a flow rate of 25 mL/min. The flow rates of the two pumps were set to the value required by the experimental design.

Experimental design

The influences of three independent variables, namely, the ethanol concentration (v/v, X₁), extraction time (h, X₂), and ratio of recycling ethanol flow rate and initial solvent volume in the extraction tank (RES, min⁻¹, X₃), were investigated. A three-variable, three-level Box–Behnken design (BBD) was employed. The coded and uncoded values of the three independent variables are given in [Table 1](#). The run order of the experiments is listed in [Table 2](#). After the design space was developed, verification experiments with the conditions listed in [Table 3](#) were carried out and repeated three times.

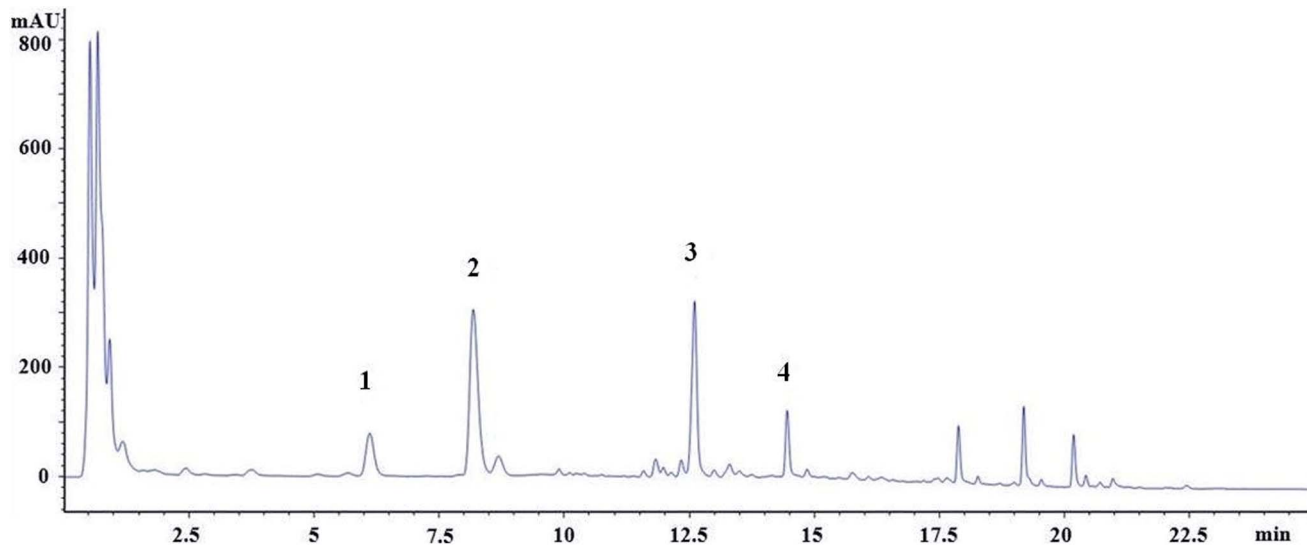


Figure 2. Chromatogram of Panax notoginseng extract.

doi:10.1371/journal.pone.0114300.g002

Analytical methods

The quantitative analysis of four saponins, namely, the notoginsenoside R₁, ginsenoside Rg₁, ginsenoside Rb₁, and ginsenoside Rd, was performed using HPLC. HPLC analysis was performed on an Agilent 1260 series HPLC system with an Acquity UPLC CSH C18 column (50 mm × 2.1 mm i.d, 1.7 μm). The column temperature was maintained at 40°C to keep column pressure in an acceptable range. The standards and samples were separated using a gradient mobile phase consisting of phase A (0.01% formic acid in deionized water) and phase B (0.01% formic acid in acetonitrile). The gradient conditions are as follows: 0–6.0 min, 18–20% B; 6.0–6.8 min, 20–30% B; 6.8–11.0 min, 30–35% B; 11.0–17.0 min, 35–90% B; and 17.0–25.0 min, 90% B. The column was then conditioned with 18% B for 15 min. The flow rate was set at 0.35 ml/min. The injection volume was 5 μL. The detection wavelength was set at 203 nm. The chromatogram is shown in [Figure 2](#). The dry matter content was determined gravimetrically using a precision electronic balance (AB204-N, Mettler Toledo Shanghai Co., Ltd.). Before weighing, the samples were dried at 105°C in an oven (DZF-6050, Shanghai Jing Hong Laboratory Instrument Co., Ltd.) for 3 h and then stored in a desiccator for 0.5 h. The pigment content was determined using spectrophotometry. The absorbance of each sample was measured at 420 nm using a UV-vis spectrophotometer (T6, Pukinje Co., Ltd., Beijing, China). The pigment yield was calculated using tartrazine as the standard.

Table 4. Upper and lower limits of the desirable ranges of CQAs.

CQA	Desirable Range	
	Low	High
R ₁ yield (mg/g <i>Panax notoginseng</i>)	13.0	15.0
Rg ₁ yield (mg/g <i>Panax notoginseng</i>)	42.0	48.0
Rb ₁ yield (mg/g <i>Panax notoginseng</i>)	43.0	51.0
Rd yield (mg/g <i>Panax notoginseng</i>)	9.0	11.0
TSP (%)	27.0	33.0
Pigment yield (equivalent of tartrazine, mg/g <i>Panax notoginseng</i>)	0.15	2.98

doi:10.1371/journal.pone.0114300.t004

Data processing

The yield of a saponin (SY) was calculated using [Equation 1](#).

$$SY_i = \frac{m_e \cdot C_i}{m_{pn}} \tag{1}$$

where C and m refer to the extract concentration and mass, respectively; subscript i (i=1 to 4) corresponds to notoginsenoside R₁, ginsenoside Rg₁, ginsenoside Rb₁, or ginsenoside Rd, respectively; and subscripts e and pn refer to the extract and *Panax notoginseng*, respectively. The concentration of total saponin (TS) in an extract was calculated using [Equation 2](#).

$$TS = \sum_{i=1}^4 C_i \tag{2}$$

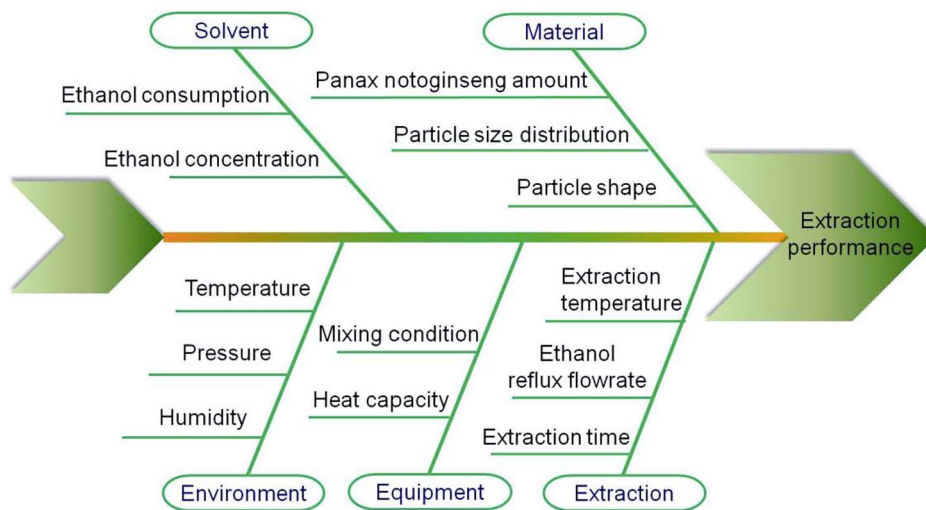


Figure 3. Ishikawa diagram analysis for the ethanol recycling extraction process.

doi:10.1371/journal.pone.0114300.g003

Table 5. Risk assessment using FMEA for the factors of ethanol recycling extraction process.

Cause	Factors	Severity ^a	Occurrence ^b	Detectability ^c	RPN
Material	<i>Panax notoginseng</i> amount	3	1	1	3
	Particle size distribution	1	2	1	2
	Particle shape	1	1	2	2
Extraction	Extraction Temperature	3	1	1	3
	Extraction Time	4	2	1	8
	Reflux rate of ethanol	3	2	2	12
Solvent	Ethanol consumption	4	1	1	4
	Ethanol concentration	4	2	2	16
Environment	Temperature	1	1	1	1
	Pressure	1	1	1	1
	Humidity	1	1	1	1
Equipment	Mix condition	2	1	1	2
	Heat capacity	2	2	1	4

^aSeverity: 1, no impact; 2, small impact; 3, moderate impact; 4, remarkable impact.

^bOccurrence: 1, seldom occur; 2, sometimes occur.

^cDetection: 1, can be detected easily; 2, can be detected with difficulty.

doi:10.1371/journal.pone.0114300.t005

The purity of total saponin (TSP) was calculated using [Equation 3](#).

$$TSP = \frac{TS}{DM} \times 100\% \tag{3}$$

where DM is the dry matter content of an extract. The yield of pigment (PY) was calculated using [Equation 4](#).

$$PY = \frac{C_{pi} \times m_e}{m_{pn}} \tag{4}$$

where the subscript pi represents the pigment content using tartrazine as the standard.

[Equation 5](#) was used to model the results of the Box-Behnken design experiments.

$$Y = a_0 + a_1A + a_2B + a_3C + a_4AB + a_5AC + a_6BC + a_7A^2 + a_8B^2 + a_9C^2 \tag{5}$$

where Y is the response, a₀ is a constant, and a₁ to a₉ are regression coefficients. Logarithmic values were used in the model development for pigment yield. Design Expert V8.0.6.1 (State-Ease Inc., MN) was used to analyze the results of the Box-Behnken design experiments.

A self-written Matlab (R2010b, Version 7.11, MathWorks, USA) program was used to calculate the design space using a Monte-Carlo method. In the Monte-Carlo simulation, it is hypothesized that the relative standard deviations (RSD) of concentrations for all of the experimental results were the same as the RSD values of the center point. In each simulation, random data following a normal distribution were created. The simulation was carried out 50000 times to calculate

Table 6. Model fitting results and ANOVA results.

Model terms	Saponins yield (mg/g <i>Panax notoginseng</i>)						TSP (%)			Pigment yield (ln(μ g/g <i>Panax notoginseng</i>))		
	R ₁		R _{g1}		R _{b1}		R _d		Estimate		Prob> t	
	Estimate	Prob> t	Estimate	Prob> t	Estimate	Prob> t	Estimate	Prob> t	Estimate	Prob> t	Estimate	Prob> t
Constant	13.6768	—	46.2633	—	46.2281	—	9.4700	—	27.3738	—	8.0307	—
X ₁	-0.6278	0.0651	-2.9012	0.0087**	-4.6364	0.0016**	-0.8498	0.0006**	1.2403	0.0211*	-0.5855	0.0002**
X ₂	0.3367	0.2723	0.8127	0.3251	1.9695	0.0612	0.3506	0.0360*	0.2196	0.6027	0.3729	0.0017**
X ₃	0.9555	0.0140*	2.1364	0.0305*	3.0673	0.0116*	0.6434	0.0026**	-1.1593	0.0274*	1.0467	<0.0001**
X ₁ X ₂	0.2365	0.5702	1.6575	0.1732	0.6454	0.6134	-0.0487	0.8002	0.5534	0.3655	0.2267	0.0604
X ₁ X ₃	-0.3715	0.3821	0.6669	0.5569	0.4740	0.7092	-0.0805	0.6777	1.6996	0.0238*	0.0399	0.6985
X ₂ X ₃	0.5233	0.2324	-0.2902	0.7958	0.1268	0.9201	-0.1566	0.4280	-0.9062	0.1601	0.0765	0.4653
X ₁ ²	-1.7090	0.0049**	-5.9570	0.0014**	-6.2568	0.0021**	-1.0188	0.0015**	-1.0889	0.1024	-0.1484	0.1816
X ₂ ²	-0.7323	0.1124	-1.7398	0.1559	-1.2746	0.3334	-0.0591	0.7594	-1.3371	0.0559	0.1414	0.2000
X ₃ ²	-0.4902	0.2598	-1.0717	0.3563	-1.7609	0.1964	-0.0353	0.8544	2.9760	0.0019**	-1.4191	<0.0001**
Model P value	0.0315*		0.0151*		0.0081**		0.0036**		0.0115*		<0.0001**	
R ²	0.8827		0.9105		0.9283		0.9463		0.9188		0.9891	

*p<0.05

**p<0.01.

doi:10.1371/journal.pone.0114300.t006

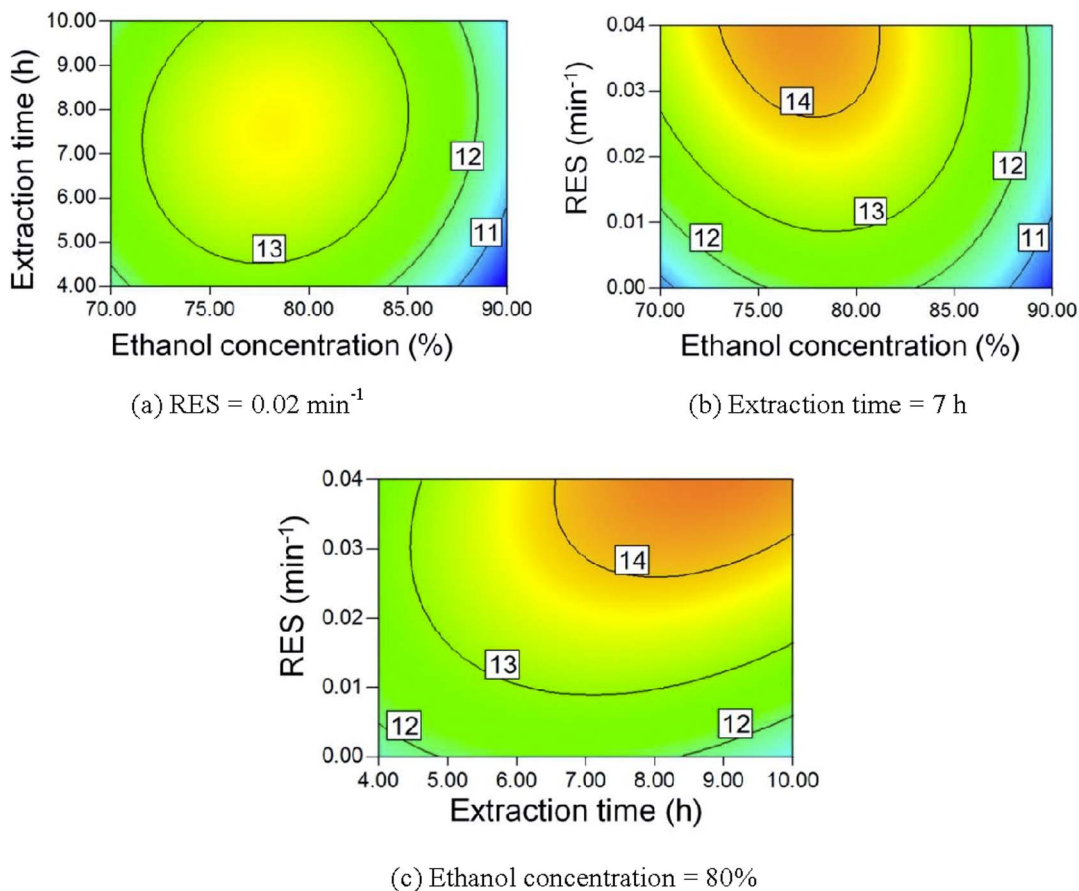


Figure 4. Contour plot of R₁ yield.

doi:10.1371/journal.pone.0114300.g004

the probability. The acceptable level of probability for the design space was set as 0.90.

Results and Discussion

Process CQA definition

Saponins are the main bioactive components of *Panax notoginseng* [18]. Saponins possess many pharmacological activities, such as antithrombotic, anti-atherosclerotic, fibrinolytic, antioxidant and cardioprotective activities [19–21].

Therefore, saponins are usually used as the main indices for *Panax notoginseng* product evaluation [22, 23]. Ginsenoside Rg₁, ginsenoside Rb₁, ginsenoside Rd, and notoginsenoside R₁ are the *Panax notoginseng* saponins present in the highest levels [24]. Recently, the action mechanisms of ginsenoside Rg₁, ginsenoside Rb₁, ginsenoside Rd, and notoginsenoside R₁ have been found to involve multiple targets and multiple pathways using a network-based approach [25]. Therefore, the yields of notoginsenoside R₁, ginsenoside Rg₁, ginsenoside Rb₁, and

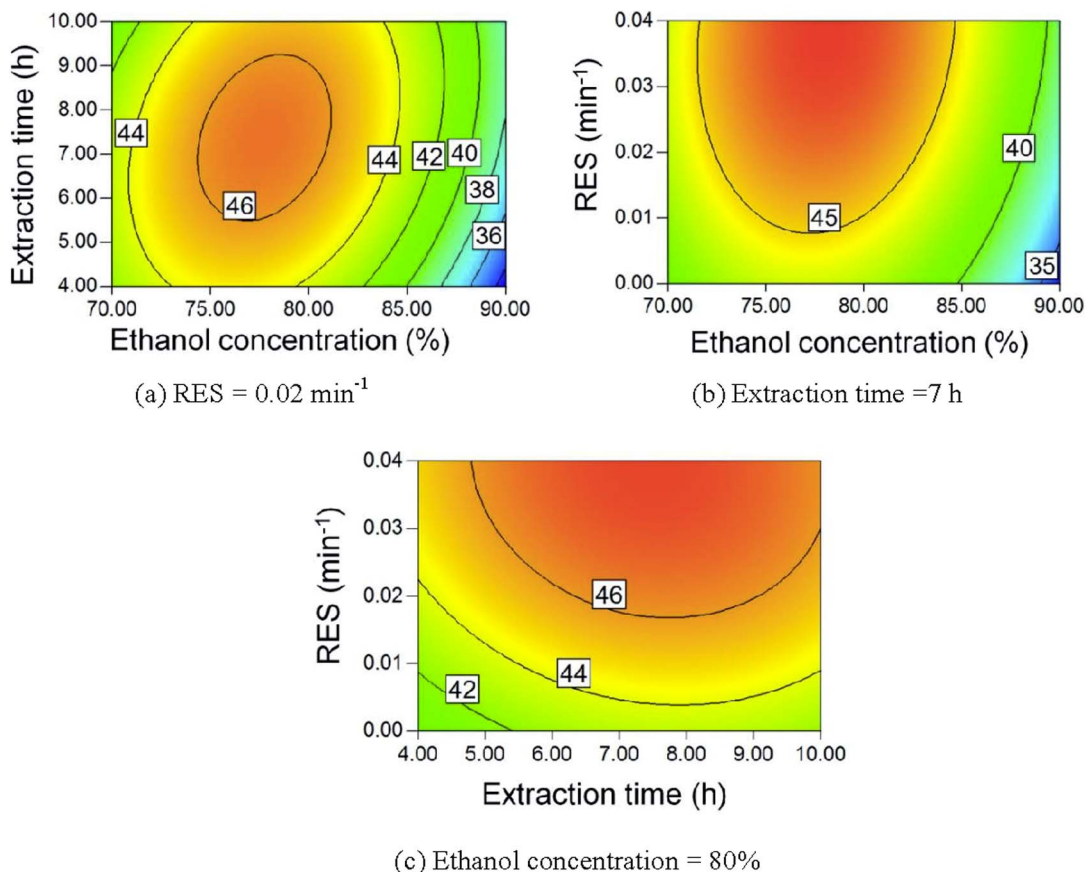


Figure 5. Contour plot of Rg₁ yield.

doi:10.1371/journal.pone.0114300.g005

ginsenoside Rd are selected as process CQAs. Higher saponin yields are favored. Considering that the active compound purity represents the difficulties in process quality control [26], total saponin purity is considered as a process CQA. In the extraction process, polysaccharides, pigments, salts, and other compounds will also be extracted, which will result in an increase in the dry matter yield. The dry matter yield is a function of saponin yield and total saponin purity, as seen in Equation 6.

$$DM = \frac{m_{pn} \times \sum_i^4 SY_i}{m_e \times TSP} \tag{6}$$

Therefore, the dry matter yield is not defined as a CQA. Color is another index of drug quality; therefore, pigment yield is also selected as a process CQA. A total of six process CQAs were taken into consideration, including the four saponin yields, total saponins purity, and pigment yield. The upper and lower limits of CQAs are given in Table 4.

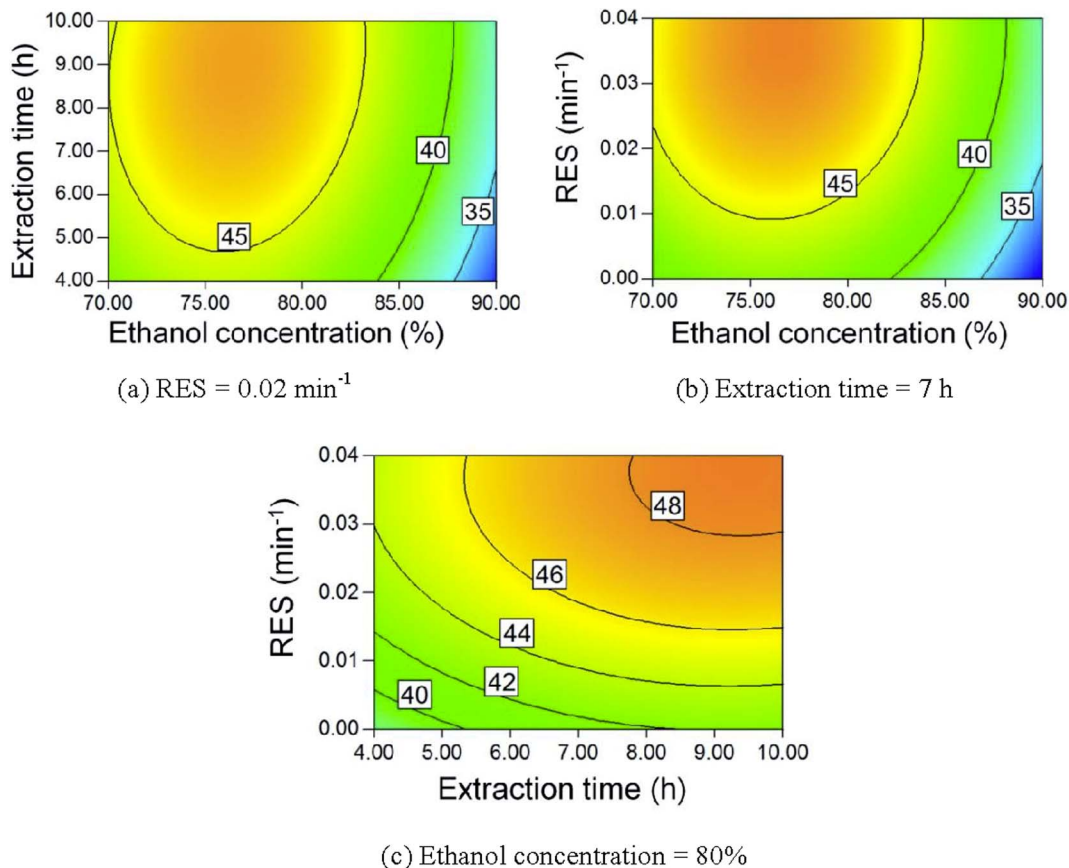


Figure 6. Contour plot of Rb₁ yield.

doi:10.1371/journal.pone.0114300.g006

CPP identification

Possible critical process parameters in the ethanol recycling reflux extraction process were identified using an Ishikawa diagram analysis, as shown in Figure 3. The main causes of environment, material attributes, solvent, equipment, and extraction procedure and the related sub-causes were considered.

For the further selection of critical process parameters (CPPs), a failure mode and effects analysis (FMEA) was conducted. In this analysis, a severity ranking from 1 to 4 is used to reflect the impact of each parameter on the process. The occurrence probability of a failure was ranked from 1 to 2 in each parameter. The ability to detect a failure was also ranked from 1 to 2 in each parameter. The scores for severity, occurrence, and detectability are obtained based on literature results and experience and are presented in Table 5. In Bai et al.'s work, the effect of the size distribution of *Panax notoginseng* on saponin extraction was small [27]. Instead, ethanol content, ethanol addition amount, and number of extractions have been found to be significant factors [24, 28]. The risk priority number (RPN) score was used as the criterion to identify the CPPs. The RPN was obtained by multiplying the scores for severity, occurrence, and detectability, as shown in

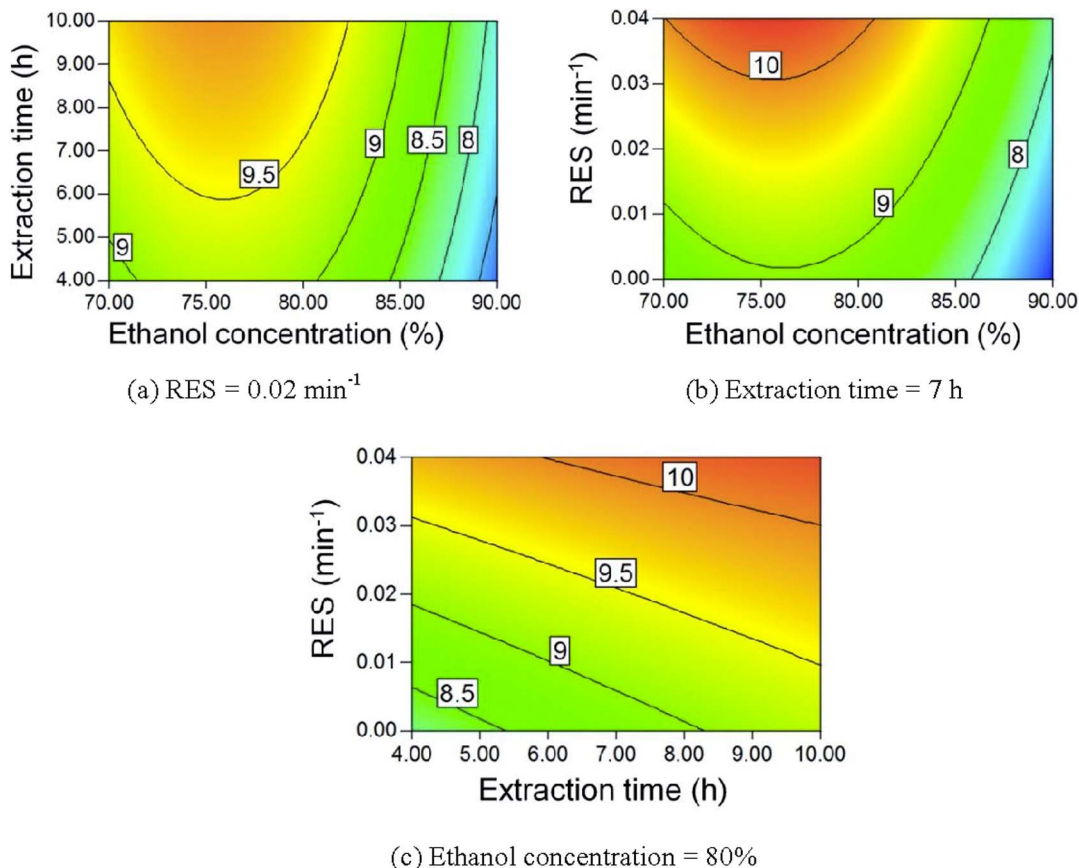


Figure 7. Contour plot of Rd yield.

doi:10.1371/journal.pone.0114300.g007

Table 5. The parameters with an RPN score of 8 or above were selected for this study. To facilitate the transfer from the laboratory scale to a larger scale, the ratio of the recycling ethanol flow rate and initial solvent volume in the extraction tank (RES) were used to represent the effects of ethanol reflux flow rate. Therefore, the ethanol content, extraction time, and RES were identified as the CPPs.

Effects of CPPs on process CQAs

The results of the Box-Behnken design experiments are shown in Table 2. The extraction yield of ginsenoside Rg₁ varied from 33.17 to 48.14 mg/g *Panax notoginseng*. The ginsenoside Rb₁ yield ranged from 30.42 to 50.71 mg/g *Panax notoginseng*. The yields of notoginsenoside R₁ and ginsenoside Rd were lower than 15 mg/g *Panax notoginseng*. The TSP varied from 23.11% to 32.55%. The pigment yield was between 0.14 and 7.87 mg/g *Panax notoginseng*.

Mathematical models were developed to describe the relationships between CQAs and CPPs. The estimated regression coefficients are listed in Table 6. Analysis of variance (ANOVA) was carried out, and the p-values of the parameters are listed in Table 6. For all of the models, the determination coefficients (R²) are

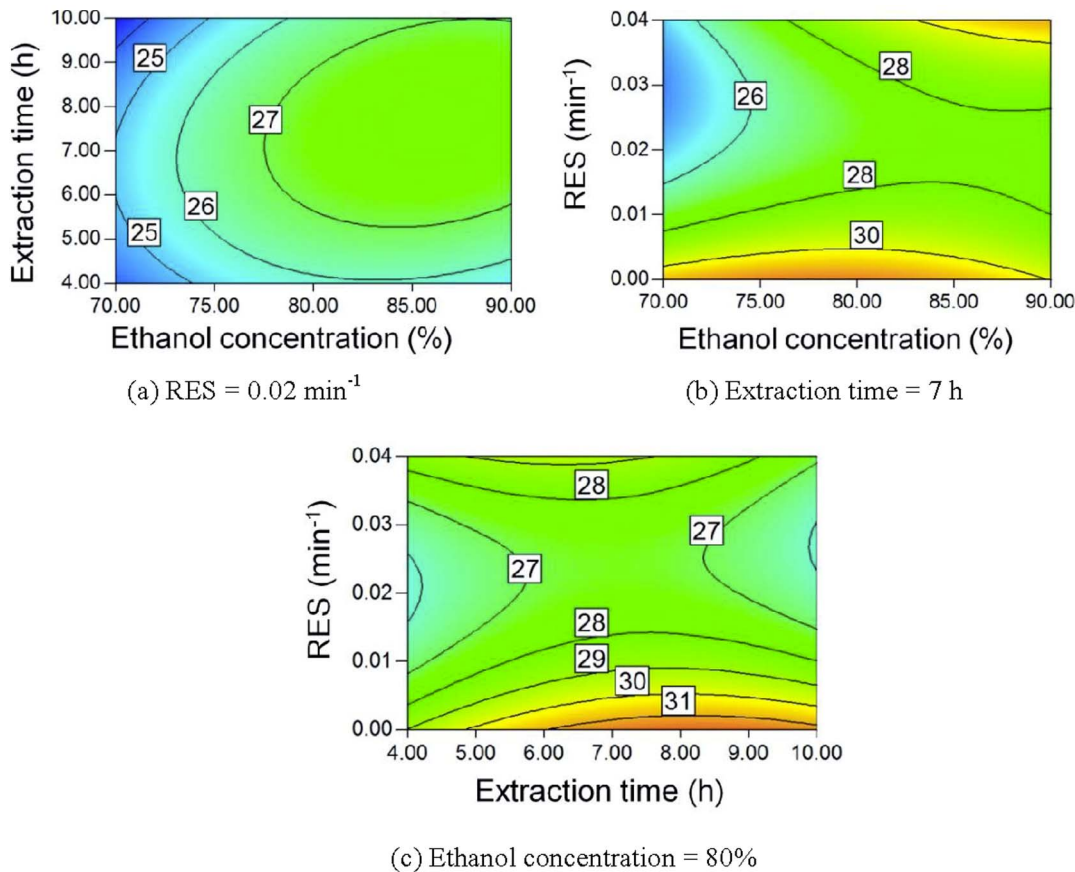


Figure 8. Contour plot of TSP.

doi:10.1371/journal.pone.0114300.g008

higher than 0.88, which means that most of the variations of the process CQAs can be explained by ethanol concentration, RES, and extraction time. The models are significant, as the p-values are less than 0.05. The linear term or quadratic term of ethanol concentration is important in all the models. The RES is also important for all of the criteria. Extraction time is significant for Rd yield and pigment yield.

According to the models, contour plots for saponin yields can be obtained, as given in Figures 4–7. As the ethanol concentration increases, the yields of all of the saponins first increase and then decrease. A lower ethanol content means more water in the extract, which can result in increased swelling of *Panax notoginseng*. Saponins can be extracted in a shorter time when the swelling of *Panax notoginseng* is greater. However, more saponins may hydrolyze when the water content in the mixed solvent is higher because of the higher boiling temperature [16, 18, 29, 30]. More ginsenoside Rb₁ and ginsenoside Rd can be extracted using longer extraction times. However, the yields of notoginsenoside R₁ and ginsenoside Rg₁ may decrease for very long extraction times. Ginsenoside Rg₁ is the hydrolysis product of notoginsenoside R₁ [16]. The hydrolyzation of ginsenoside Rg₁ forms ginsenoside Rh₁ [16]. A higher RES leads to higher

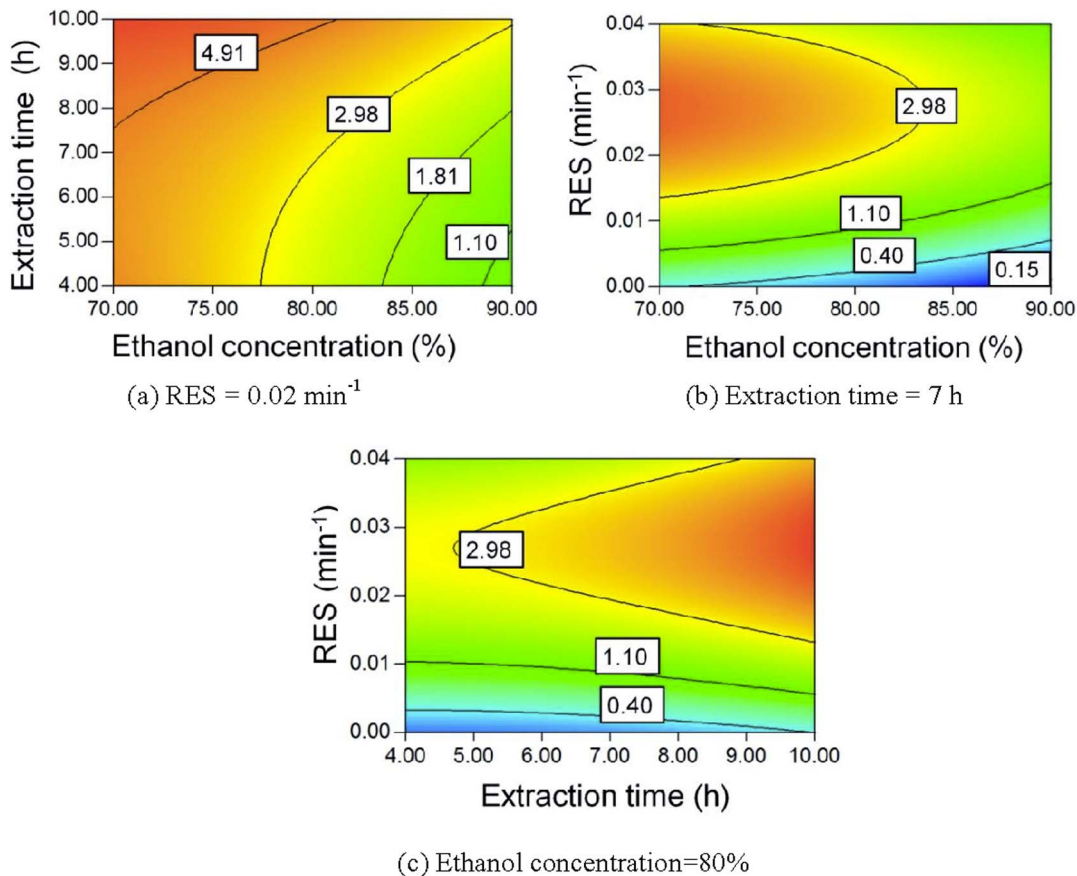


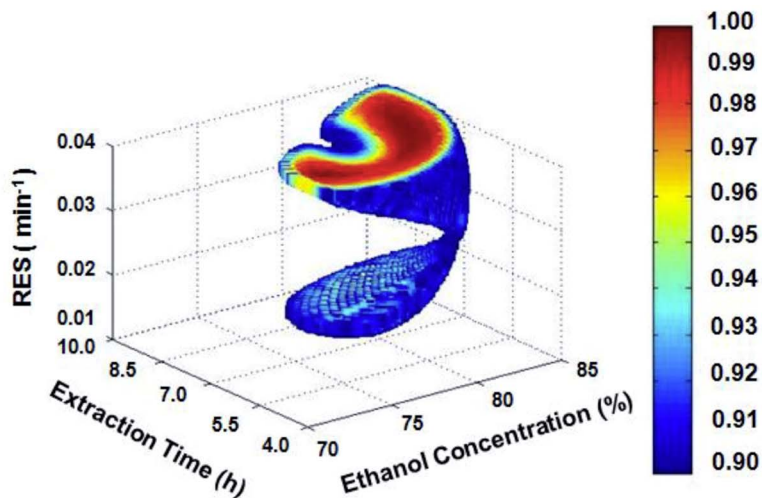
Figure 9. Contour plot of pigment yield.

doi:10.1371/journal.pone.0114300.g009

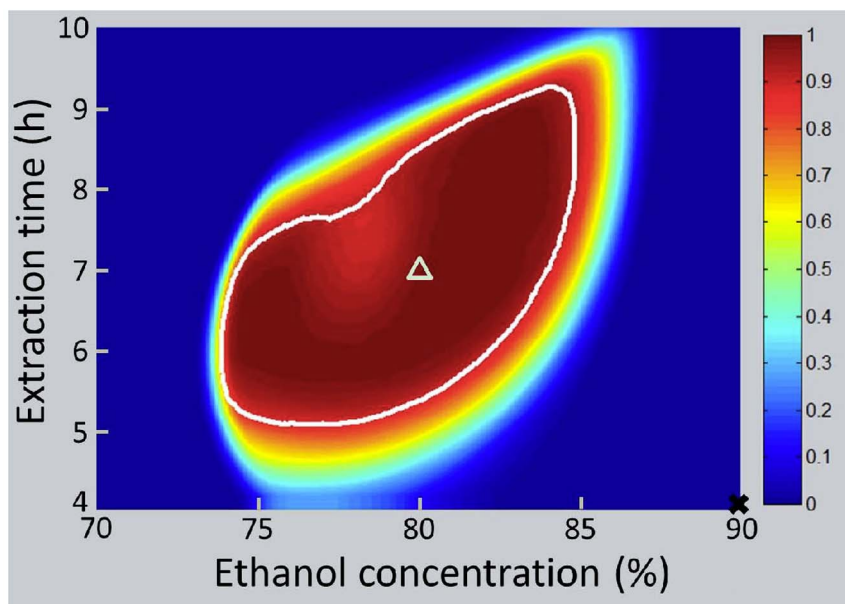
concentration differences between *Panax notoginseng* and the extract for a saponin, which accelerates the saponin extraction. Therefore, the saponin yields increase as the RES increases.

The mass of saccharides, such as sucrose, fructose, and sucrose [31], was more than 60% of the dry matter content in the water extract of *Panax notoginseng* [32]. Saccharide solubilities usually decrease as the ethanol content in the mixed ethanol-water mixture increases [33–36]. Therefore, a higher ethanol concentration means that less saccharides will be extracted and a higher TSP can be obtained, as seen in Figure 8. The extract is easily saturated by saccharides, but the recycling of the ethanol solution can prevent this saturation, thereby allowing more saccharides to be extracted. Accordingly, a lower RES also results in a higher TSP.

Maillard reaction between reducing sugars and amino acids forms aroma compounds, ultra-violet-absorbing intermediates, and melanoidins, which results in a darkened extract [37, 38]. Longer extraction times result in a larger amount of Maillard reaction products. Accordingly, the pigment yield increases, as seen in Figure 9(a) and 9(c). A lower ethanol concentration corresponds to a higher water



(a) Three-dimensional design space



(b) RES = 0.04 min⁻¹; Δ, V₁; ×, V₂; the region enclosed by the white line is the design space

Figure 10. Design space and verification experiments.

doi:10.1371/journal.pone.0114300.g010

content in the extract and a higher extraction temperature for reflux. Water is also a reactant in the Maillard reaction [39]. Therefore, a lower ethanol concentration also leads to a higher pigment yield, as seen in Figure 9(a) and 9(b).

Design space development and verification

The design space with a possibility above 0.90 of satisfying the CQA criteria is calculated using a Monte-Carlo method. The results are shown in Figure 10(a).

The design space is irregular in shape and composed of two parts. For easier operation, normal operation ranges are calculated, comprised of an ethanol concentration of 79–82%, extraction time of 6.1–7.1 h, and RES of 0.039–0.040 min⁻¹. The probability of satisfying the CQA criteria is 0.914 when the parameters are controlled within the normal operation region.

To further confirm the accuracy of the models as well as the design space, verification experiments were carried out. The experimental conditions and results are shown in [Table 3](#). The conditions of Experiments V₁ and V₂ are plotted in [Figure 10\(b\)](#). The relative deviation (RD) values between the experimental and predicted values are calculated using [Equation 7](#).

$$RD = \frac{|V_{Exp} - V_{Cal}|}{V_{Exp}} \times 100\% \quad (7)$$

where V_{Exp} is the average value of the verification experimental data, and V_{Cal} is the calculated value of the model. The RD values are less than 6% for ginsenoside Rg₁ yield, ginsenoside Rb₁ yield, ginsenoside Rd yield, and TSP of the extract, indicating that these models are accurate. The results of Experiment V₁ are within the target ranges of the CQAs. However, the yields of the four different saponins in Experiment V₂ are lower than their criteria. Thus, operation within the design space can ensure that all CQAs are within the predefined limits.

Conclusion

The design space approach is applied to optimize the solvent recycling reflux extraction process of *Panax notoginseng*. The notoginsenoside R₁ yield, ginsenoside Rg₁ yield, ginsenoside Rb₁ yield, ginsenoside Rd yield, pigment yield, and total saponin purity in the extracts were defined as the process CQAs. An Ishikawa diagram was applied to identify potential CPPs. Ethanol concentration, extraction time, and RES were identified as CPPs using an FMEA model. The models between the CPPs and process CQAs were built using quadratic models with R² values greater than 0.88. As the ethanol concentration increases, the saponin yields first increase and then decrease. For the ginsenosides Rb₁ and Rd, a longer extraction time leads to higher yields. The total saponin purity increases as the ethanol concentration increases. Meanwhile, the pigment yield decreases with increasing ethanol concentration or decreasing extraction time. The design space was calculated via Monte-Carlo simulation using 0.90 as the acceptable probability. Normal operation ranges were also recommended, namely, an ethanol concentration of 79–82%, extraction time of 6.1–7.1 h, and RES of 0.039–0.040 min⁻¹. The probability of satisfying the CQA criteria with parameters in normal operation ranges is higher than 91.4%. The design space was verified, and the results of the verification experiment showed that the use of operating CPPs within the design space provides a high probability of satisfying the process CQA criteria.

Author Contributions

Conceived and designed the experiments: HBQ XCG. Performed the experiments: YZ JYP. Analyzed the data: XCG HBQ YZ JYP. Contributed reagents/materials/analysis tools: HBQ XCG. Wrote the paper: XCG HBQ YZ JYP.

References

1. **Chen JY, Liu Z, Wu M, Ke YC** (2013) Optimization of extraction process of Xinmaikang tablets by orthogonal test. *Chinese Journal of Experimental Traditional Medical Formulae* 19: 59–61.
2. **Li ZX, Men LH, He KD** (2009) Study and comparison of double vessels double flow directions external cyclic heat feedback extracting and normal extracting techniques. *Pharmaceutical & Engineering Design* 30: 35–38.
3. **Zhao YL, Hou QC, Li L, Guo JR, Cui YJ, et al.** (2011) Preparation of polysaccharides from *Grifola frondosa* using dynamic hot reflux extraction. *Journal of Central South University (Science and Technology)* 42: 38–45.
4. **Wu M, Chen JY, Dong MG, LIU Z** (2013) Process research on the preparation technology of Jianweixiaozhang pill. *Journal of Jiangxi University of TCM* 25: 66–68.
5. **Yu LX** (2008) Pharmaceutical quality by design: Product and process development, understanding, and control. *Pharm Res-Dordr* 25: 781–791.
6. **Gong XC, Wang SS, Li Y, Qu HB** (2013) Separation characteristics of ethanol precipitation for the purification of the water extract of medicinal plants. *Sep Purif Technol* 107: 273–280.
7. **International Conference on Harmonization (ICH)** (2009/4/) Pharmaceutical development. Q8(R2). http://www.ich.org/fileadmin/Public_Web_Site/ICH_Products/Guidelines/Quality/Q8_R1/Step4/Q8_R2_Guideline.pdf.
8. **Zhang L, Yan BJ, Gong XC, Yu LX, Qu HB** (2013) Application of Quality by Design to the Process Development of Botanical Drug Products: A Case Study. *Aaps Pharmscitech* 14: 277–286.
9. **Debrus B, Lebrun P, Ceccato A, Caliaro G, Rozet E, et al.** (2011) Application of new methodologies based on design of experiments, independent component analysis and design space for robust optimization in liquid chromatography. *Anal Chim Acta* 691: 33–42.
10. **Rozet E, Lebrun P, Debrus B, Boulanger B, Hubert P** (2013) Design Spaces for analytical methods. *Trac-Trend Anal Chem* 42: 157–167.
11. **Peterson JJ** (2008) A Bayesian approach to the ICH Q8 definition of design space. *J Biopharm Stat* 18: 959–975.
12. **Nistor I, Lebrun P, Ceccato A, Lecomte F, Slama I, et al.** (2013) Implementation of a design space approach for enantiomeric separations in polar organic solvent chromatography. *J Pharmaceut Biomed* 74: 273–283.
13. **Rozet E, Lebrun P, Debrus B, Hubert P** (2012) New methodology for the development of chromatographic methods with bioanalytical application. *Bioanalysis* 4: 755–758.
14. **Gong X, Li Y, Guo Z, Qu H** (2014) Control the effects caused by noise parameter fluctuations to improve pharmaceutical process robustness: A case study of design space development for an ethanol precipitation process. *Sep Purif Technol* 132: 126–137.
15. **Gong X, Chen H, Chen T, Qu H** (2014) Unit Operation Optimization for the Manufacturing of Botanical Injections Using a Design Space Approach: A Case Study of Water Precipitation. *PloS one* 9: e104493.
16. **Wang D, Liao PY, Zhu HT, Chen KK, Xu M, et al.** (2012) The processing of *Panax notoginseng* and the transformation of its saponin components. *Food Chem* 132: 1808–1813.
17. **Lebrun P, Krier F, Mantanus J, Grohganz H, Yang MS, et al.** (2012) Design space approach in the optimization of the spray-drying process. *Eur J Pharm Biopharm* 80: 226–234.
18. **Sun S, Wang CZ, Tong R, Li XL, Fishbein A, et al.** (2010) Effects of steaming the root of *Panax notoginseng* on chemical composition and anticancer activities. *Food Chem* 118: 307–314.

19. **Wan JB, Yang FQ, Li SP, Wang YT, Cui XM** (2006) Chemical characteristics for different parts of Panax notoginseng using pressurized liquid extraction and HPLC-ELSD. *Journal of pharmaceutical and biomedical analysis* 41: 1596–1601.
20. **Li SP, Qiao CF, Chen YW, Zhao J, Cui XM, et al.** (2013) A novel strategy with standardized reference extract qualification and single compound quantitative evaluation for quality control of Panax notoginseng used as a functional food. *Journal of chromatography A* 1313: 302–307.
21. **Zhu J, Fan X, Cheng Y, Agarwal R, Moore CM, et al.** (2014) Chemometric Analysis for Identification of Botanical Raw Materials for Pharmaceutical Use: A Case Study Using Panax notoginseng. *PLOS ONE* 9: e87462.
22. **Wang YT, You JY, Yu Y, Qu CF, Zhang HR, et al.** (2008) Analysis of ginsenosides in Panax ginseng in high pressure microwave-assisted extraction. *Food Chem* 110: 161–167.
23. **Li SX, Mu Y, Zheng FY** (2013) Influence of Gastrointestinal Digestion and Edible Plant Combination on Oral Bioavailability of Triterpene Saponins, Using a Biomimetic Digestion and Absorption System and Determination by HPLC. *J Agr Food Chem* 61: 10599–10603.
24. **Dong TT, Zhao KJ, Huang WZ, Leung KW, Tsim KW** (2005) Orthogonal array design in optimizing the extraction efficiency of active constituents from roots of Panax notoginseng. *Phytotherapy research: PTR* 19: 684–688.
25. **Wang L, Li Z, Zhao X, Liu W, Liu Y, et al.** (2013) A network study of chinese medicine xuesaitong injection to elucidate a complex mode of action with multicomponent, multitarget, and multipathway. *Evidence-based complementary and alternative medicine: eCAM* 2013: 652373.
26. **Gong XC, Wang SS, Qu HB** (2011) Comparison of Two Separation Technologies Applied in the Manufacture of Botanical Injections: Second Ethanol Precipitation and Solvent Extraction. *Ind Eng Chem Res* 50: 7542–7548.
27. **Bai C, Chai X, Gao X, Li P, Tu P** (2009) [HPLC-CAD in optimizing of saponins extraction from Radix et Rhizoma Notoginseng]. *Zhongguo Zhong yao za zhi = Zhongguo zhongyao zazhi = China journal of Chinese materia medica* 34: 677–680.
28. **Qu LH, Zheng M, Lou YJ** (2006) [Studies on extraction process optimization of Panax notoginseng saponins]. *Zhong yao cai = Zhongyaocai = Journal of Chinese medicinal materials* 29: 593–595.
29. **Sun S, Qi LW, Du GJ, Mehendale SR, Wang CZ, et al.** (2011) Red notoginseng: higher ginsenoside content and stronger anticancer potential than Asian and American ginseng. *Food Chem* 125: 1299–1305.
30. **Wu W, Qin Q, Guo Y, Sun J, Liu S** (2012) Studies on the chemical transformation of 20(S)-protopanaxatriol (PPT)-type ginsenosides R(e), R(g2), and R(f) using rapid resolution liquid chromatography coupled with quadruple-time-of-flight mass spectrometry (RRLC-Q-TOF-MS). *J Agric Food Chem* 60: 10007–10014.
31. **Yang G, Tao N, Hong C, Bo X, Ling-yi K** (2012) Determination of Monosaccharide and Disaccharide in Panax notoginseng by HPLC with SPE and ELSD. *Chin Med J Res Prac* 26: 29–31.
32. **Liu G, Liu Y, Ban J, Zhang C** (2005) Analysed the contents of sugar constituents in Panax notoginseng (Burk) F. H. Chen and rhizoma of Panax notoginseng(Burk) F. H. Chen. *RENSHEN YANJ IU*: 34–36.
33. **Gong XC, Wang SS, Qu HB** (2011) Solid-Liquid Equilibria of D-Glucose, D-Fructose and Sucrose in the Mixture of Ethanol and Water from 273.2 K to 293.2 K. *Chinese J Chem Eng* 19: 217–222.
34. **Bouchard A, Hofland GW, Witkamp GJ** (2007) Properties of sugar, polyol, and polysaccharide water-ethanol solutions. *J Chem Eng Data* 52: 1838–1842.
35. **Tsavas P, Voutsas E, Magoulas K, Tassios D** (2004) Phase equilibrium calculations in aqueous and nonaqueous mixtures of sugars and sugar derivatives with a group-contribution model. *Ind Eng Chem Res* 43: 8391–8399.
36. **Gong XC, Wang C, Zhang L, Qu HB** (2012) Solubility of Xylose, Mannose, Maltose Monohydrate, and Trehalose Dihydrate in Ethanol-Water Solutions. *J Chem Eng Data* 57: 3264–3269.
37. **Kim J-S, Lee Y-S** (2008) Effect of reaction pH on enolization and racemization reactions of glucose and fructose on heating with amino acid enantiomers and formation of melanoidins as result of the Maillard reaction. *Food Chem* 108: 582–592.
38. **Coca M, Garcia MT, Gonzalez G, Pena M, Garcia JA** (2004) Study of coloured components formed in sugar beet processing. *Food Chem* 86: 421–433.

39. **Davidek T, Robert F, Devaud S, Vera FA, Blank I** (2006) Sugar fragmentation in the Maillard reaction cascade: Formation of short-chain carboxylic acids by a new oxidative α -dicarbonyl cleavage pathway. *J Agr Food Chem* 54: 6677–6684.